

Chapter 4 Test
The Periodic Table of Elements

Part 1- 30 Multiple Choice Questions (2 pts each)

Directions: Choose the letter that best answers the question or completes the statement.

1. The ability to pound a metal into a flat sheet without breaking it is called:
A. luster C. ductility
B. malleability D. conductivity
2. All of the following are physical properties of metals except:
A. luster C. corrosion
B. malleability D. low melting points
3. Which class of elements best conducts electricity?
A. metals C. metalloids
B. nonmetals D. noble gases
4. Elements on the periodic table are ordered by:
A. atomic number C. atomic symbol
B. atomic mass D. electron shells
5. An element in Group 2 on the Periodic table would be considered a(n):
A. Transition metal C. Alkaline earth metal
B. Metalloid D. Halogen
6. What do the groups/families in the periodic table have in common?
A. the number of electron shells
B. the state of matter
C. the amount of isotopes
D. similar physical and chemical properties
7. What do the elements sulfur (S), nitrogen (N), helium (He), and carbon (C) have in common?
A. they are noble gases
B. they are nonmetals
C. they have the same number of neutrons
D. they have the same number of protons
8. Boron's atomic number is 5. This means that a boron atom has:
A. five neutrons in its nucleus
B. five protons in its nucleus
C. a total of five protons and neutrons
D. a total of five neutrons and electrons
9. Hydrogen has three naturally occurring isotopes. Which of the following describes the relationship of the isotopes?
A. different mass, different atomic number
B. same mass, different atomic number
C. different mass, same atomic number
D. same mass, same atomic number

Elements And The Periodic Table Chapter Test Answer Key

Harold L. Levin, David T. King, Jr.



Elements And The Periodic Table Chapter Test Answer Key:

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