

Lewis Dot Structures

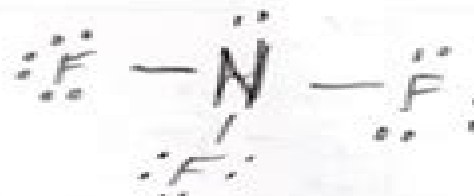
Name _____ Period _____

Write the formula where appropriate and draw Lewis dot structures for the following Compounds.

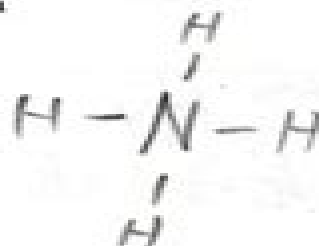
dihydrogen monoxide



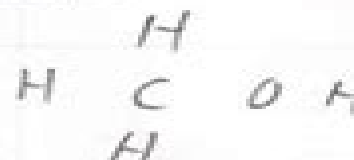
Nitrogen trifluoride



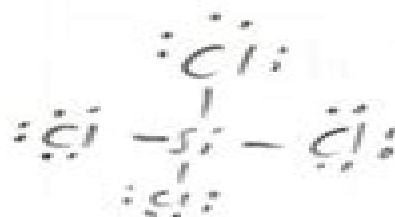
Ammonia



CH₃OH



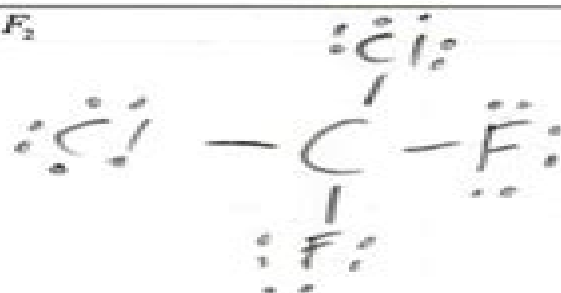
Silicon tetrachlorine



Chlorine monofluoride



CCl₃F₂



HOCl



Electron Dot Diagrams And Lewis Structures Answers

Lingsheng Yao



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basics before they progress to an advanced general chemistry course Enhanced with visualization techniques such as the first chapter s mythical microscope the book clarifies challenging abstract ideas and stimulates curiosity into what can otherwise be an overwhelming topic Topics discussed in this reader friendly text include Properties and structure of matter Atoms molecules and compounds The Periodic Table Atomic weight formula weights and moles Gases and solutions Chemical equilibrium Acids bases and pH Organic chemicals The appendix contains answers to the homework exercises so students can check their work and receive instant feedback as to whether they have adequately grasped the concepts before moving on to the next section Designed to help students embrace chemistry not with trepidation but with confidence this solid preparatory text forms a firm foundation for more advanced chemistry training

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principles of chemistry including formulas and names chemical bonds

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