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Electrolysis

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- In an electrolysis cell, the anode is the positive electrode and the cathode is the negative electrode.
- In an electrolysis cell, the anode is the positive electrode and the cathode is the negative electrode.
- Electrolysis is a process in which electricity is used to drive a chemical reaction that breaks down a substance into its constituent elements.

Electrolysis of Molten Sodium Chloride

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Electrolysis



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