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web one of the ions produced is always h which immediately combines with a water molecule to form the hydronium ion h₃o⁺ the h₃o⁺ ion is what defines the acidic properties of a substance because of their special classification acids have a naming system different from ionic or other molecular covalent compounds

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web the molecular or empirical formula acids have their own set of naming rules cf sec 2 8 which you must also know you
also need to be able to recognize and name a few organic compounds such as simple alkanes and alcohols see sec 2 9 key
questions 8 in the chemical formula and name which element is given first a metal or nonmetal 9

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web each pair has one acid that ends in ic and another that ends in ous these endings are related to the name of the
polyatomic anion found in the acid ate or ite complete the statements below with the correct acid name ending polyatomic
anion ending is ate acid name ending is if