#### Notes Lewis Dot Structures

Representative of only the valence electrons

<u>Valence electrons</u>: electrons in the outermost energy level.

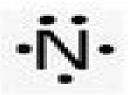
# Periodic Table Label: 12345678

Ex Na

·Na

Exc. N

Ex: S





# **Notes Bonding**

Chemical Formula: shows the lands and numbers of allows in the smallest representative and of a substance.

SAREGREE		Suprem ceight	
Cl <sub>2</sub>	Tells the number of elements present 2 CFs	CI1-	Tells the charge of the ion - means a GAH of electrons - anion
H₂O H₃PO₄	H: 2 O: 1 H: 3 F: 1 O: 4	Mg <sup>2+</sup> CO <sub>3</sub> <sup>2-</sup>	+ Means a LOSS of electrons - cohon A charge on multiple elements means it is a potratomic long the charge is shared over the entire compound.

# **Guided Notes 8 The Covalent Bond Answer Key**

**Arshad Iqbal** 

#### **Guided Notes 8 The Covalent Bond Answer Key:**

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