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Released Question from Chemistry EOC

5 Element X has two known naturally occurring isotopes. The mass and relative abundance of each isotope are shown below.

| Relative Abundance | Mass (amu) |
|--------------------|------------|
| 50.57% | 78.92 |
| 49.43% | 80.92 |

What is the average atomic mass of Element X to the nearest hundredth of an atomic mass unit?

Record your answer and fill in the bubbles on your answer document.

Step 1: Convert % to Decimals.

Step 2: Multiple Relative Abundance by Mass.

Step 3: Add the numbers for final answer.

Step 4: Round correctly to the nearest hundredth and grid answer.

Eoc Release Questions For Chemistry

L Cohen

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Decoding **Eoc Release Questions For Chemistry**: Revealing the Captivating Potential of Verbal Expression

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