

**Chapter 6 Test Periodic Table****Matching**

*Match each item with the correct statement below.*

- |                 |                     |
|-----------------|---------------------|
| a. period       | d. metal            |
| b. periodic law | e. transition metal |
| c. group        |                     |

- \_\_\_\_\_ 1. horizontal row in the periodic table
- \_\_\_\_\_ 2. vertical column in the periodic table
- \_\_\_\_\_ 3. A repetition of properties occurs when elements are arranged in order of increasing atomic number.
- \_\_\_\_\_ 4. type of element that is a good conductor of heat and electric current
- \_\_\_\_\_ 5. type of element characterized by the presence of electrons in the *d* orbital

**Multiple Choice**

*Identify the choice that best completes the statement or answers the question.*

- \_\_\_\_\_ 6. What is another name for the representative elements?
- |                     |                        |
|---------------------|------------------------|
| a. Group A elements | c. Group C elements    |
| b. Group B elements | d. transition elements |
- \_\_\_\_\_ 7. What is another name for the transition metals?
- |                     |                     |
|---------------------|---------------------|
| a. noble gases      | c. Group B elements |
| b. Group A elements | d. Group C elements |
- \_\_\_\_\_ 8. Which of the following elements is in the same period as phosphorus?
- |           |              |             |           |
|-----------|--------------|-------------|-----------|
| a. carbon | b. magnesium | c. nitrogen | d. oxygen |
|-----------|--------------|-------------|-----------|
- \_\_\_\_\_ 9. Each period in the periodic table corresponds to \_\_\_\_\_.
- |                             |                 |
|-----------------------------|-----------------|
| a. a principal energy level | c. an orbital   |
| b. an energy sublevel       | d. a suborbital |
- \_\_\_\_\_ 10. The modern periodic table is arranged in order of increasing atomic \_\_\_\_\_.
- |         |           |           |           |
|---------|-----------|-----------|-----------|
| a. mass | b. charge | c. number | d. radius |
|---------|-----------|-----------|-----------|
- \_\_\_\_\_ 11. Who arranged the elements according to atomic mass and used the arrangement to predict the properties of missing elements?
- |                      |                     |
|----------------------|---------------------|
| a. Henry Moseley     | c. John Dalton      |
| b. Antoine Lavoisier | d. Dmitri Mendeleev |
- \_\_\_\_\_ 12. Which of the following categories includes the majority of the elements?
- |               |            |           |              |
|---------------|------------|-----------|--------------|
| a. metalloids | b. liquids | c. metals | d. nonmetals |
|---------------|------------|-----------|--------------|
- \_\_\_\_\_ 13. Of the elements Pt, V, Li, and Kr, which is a nonmetal?
- |       |      |       |       |
|-------|------|-------|-------|
| a. Pt | b. V | c. Li | d. Kr |
|-------|------|-------|-------|

# End Of Chapter Test Chapter 9 Carnegie Learning 1183

**AW Rasmussen**



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