Let the relative atomic mass of element Z be z.

Z	Bromine, Br
2.07	3.67 - 2.07 = 1.60
2.07 z	$\frac{1.60}{80} = 0.02$

Based on the empirical formula ZBr_2 , the ratio of atoms of Z: Br is 1 : 2.

Therefore,
$$\frac{2.07}{z} : 0.02 = 1 : 2$$

$$\frac{2.07}{z} \div 0.02 = 1 \div 2$$

$$\frac{2.07}{0.02z} = \frac{1}{2}$$

$$z = \frac{2 \times 2.07}{0.02}$$

$$= 207$$

The relative atomic mass of element Z is 207.

Empirical Formula For Copper Oxide

Jacqueline Fergusson, Richard Hart

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