

IGCSE Physics Paper 6 tips!



Lenses Precautions:

- Carry out experiment in a dark room.
- Use a set square to make sure the screen is vertical.
- Measure the distance between the lens and the object by drawing a vertical line in the middle of the lens' holder, and measure the distance from there.
- Move the lens back and forth until you have a focused image.
- The centers of the lens, object, and screen must have the same height from the bench.
- Lens, object and screen must be perpendicular to the bench.
- Clamp rule or place on the bench.

Pins precautions :

- Place pins as far away from each other as possible.
- View pins with one eye such that they appear as one.
- View pins from their bases in case one of them isn't vertical.

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Mike Folland

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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web jun 23 2023 p2 lewis structure has two phosphorus atoms p which contain a triple bond between them there is 1 lone pair on both the phosphorus atoms p if you haven't understood anything from the above image of p2 lewis structure then just stick with me and you will get the detailed step by step explanation on drawing a lewis structure of p2

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web these structures also known as lewis structures or electron dot structures are drawings that visually demonstrate how electrons are shared and arranged around atoms the electrons denoted as dots are called lone pairs and belong to an individual atom electrons denoted as lines are bonds and show the sharing of two electrons between two

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web when you are trying to draw a lewis structure there are guidelines to do it number of steps can be changed according to the complexity of the molecule or ion because p2o5 molecule is a bit complex molecule if you are a beginner to lewis structure drawing you have to be careful when it starts to draw

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web dec 13 2017 step 1 draw the lewis structure there are three central atoms p p o arrange them symmetrically in the order p o p attach the remaining o atoms as terminal atoms and arrange electrons to give each atom an octet the p atoms will get expanded octets step 2 use vsepr theory to determine the electron geometries

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web jun 27 2022 a lewis electron dot diagram or electron dot diagram or a lewis diagram or a lewis structure is a representation of the valence electrons of an atom that uses dots around the symbol of the element the number of dots equals the number of valence electrons in the atom

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web a lewis structure is a way to show the shape of a molecule dots show where electrons are around the atoms and lines or pairs of dots show where covalent bonds connect the atoms by drawing a lewis dot structure you can find the lone electron pairs in molecules which helps you figure out how chemical bonds form

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web oct 8 2023 introduction lewis structure is basically a graphic representation of the electron distribution around an atom the major reason why learning lewis dot structure is important is that it helps in predicting the number and type of bonds which can be formed around an atom it also helps in predicting the geometry of the molecule