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 $\text{N}_2 + \text{F}_2 \rightarrow \text{NF}_3$ 2 $\text{C}_6\text{H}_{10}\text{O}_2 + \text{CO}_2 + \text{H}_2\text{O} \rightarrow 3\text{HBr} + \text{KHCO}_3 + \text{H}_2\text{O} + \text{KBr} + \text{CO}_2$ 3 $\text{GaBr}_3 + \text{NaBr} \rightarrow \text{SNO}_4 + \text{NaSO}_3 + \text{Ga}_2\text{SO}_3 + 3\text{SO}_2 + 3\text{NF}_3 + 3\text{SnF}_2 + \text{N}_2\text{O}_3$ using the equation from problem 2 above answer the following

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 $\text{br } 3 \text{ } 1 \text{ } \text{hbr} \text{ } 1 \text{ } \text{khco}_3 \text{ } 1 \text{ } \text{h}_2\text{o} \text{ } 1 \text{ } \text{kbr} \text{ } 1 \text{ } \text{co}_2$
 $\text{br } 4 \text{ } 2 \text{ } \text{gabr}_3 \text{ } 3 \text{ } \text{na}_2\text{so}_3 \text{ } 1 \text{ } \text{ga}_2 \text{ } \text{so}_3 \text{ } 3 \text{ } 6 \text{ } \text{nabr}$
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 br using the following equation
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 answer
 $\text{ce } \text{sio}_2 \text{ } 3\text{c} \rightarrow \text{sic} \text{ } 2\text{co}$

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$$\text{Mg(OH)}_2 + \text{HCl} \rightarrow \text{MgCl}_2 + \text{H}_2\text{O}$$
 note all reactants and products require a coefficient of at least one
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 unit 4 mass spectrometry
 unit 5 chemical reactions and stoichiometry
 unit 6 more about chemical reactions
 unit 7 electronic structure of atoms
 unit 8 periodic table