

Unit 7: Chemical Formulas and Compounds Guided Notes

I. Ch. 7.1: Naming Ionic Compounds

A. Warm Up:

1. Distinguish between a cation and an anion.
2. Which elements tend to form cations? anions?
3. Using a periodic table, predict the charge ion each of the following elements form:
 - a) calcium
 - b) fluorine
 - c) sulfur
 - d) aluminum

B. Writing Formulas and Naming Chemical Compounds

1. There are two types of chemical compounds:
 - a) ionic compounds
 - b) covalent (molecular) compounds

C. Ionic Compounds are composed of a cation and an anion.

1. cations – positively charged ion (_____)
2. anions – negatively charged ion (_____)
3. Monoatomic ions
 - a) Cations
 - (1) For monatomic cations, the name of the ion is the name of the element.
 - (2) examples:
 - (a) Na^+ = _____
 - (b) Mg^{2+} = _____
 - (c) Al^{3+} = _____
 - b) Some metals are able to form more than one stable ion (_____)
 - (1) usually transition elements
 - (2) example: _____
 - (3) forms Cu^{2+} and Cu^+
 - (4) use Roman numerals to distinguish charge
 - (5) copper (II) and copper (I)

Guided And Study Chemical Names Formulas

Liying Dong



Guided And Study Chemical Names Formulas:

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