

Chemistry EOC Review Packet

Matter and Change: Pages 34-50

Procedural:

- Chemical change
- Chemical property
- Physical change
- Physical property
- Heterogeneous mixture
- Heterogeneous mixture
- Compound
- Element
- Gas
- Liquid
- Solid
- Law of Conservation of Mass

Concept Questions:

1. Write the definition of matter. Give 3 examples of matter and 3 that are not matter.

Anything that has mass and takes up space: matter has substance + form (light, heat)

2. Fill in the chart with whether matter should be classified as solid, liquid, or gas.

State of Matter	Volume	Shape
Solid	constant	constant
Liquid	constant	variable
Gas	variable	variable

3. Which picture represents each state of matter?



Solid



Liquid



Gas

4. Give 3 properties of each of the particles.

Solid: particles vibrate in position; shape constant

5. Give 3 properties of each of the particles.

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Wolfgang Guggemos

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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