

Empirical Formula Practice 1.0 – Answer Key**Objective:** - Calculate the empirical formula of a compound given experimental data**Directions:** Answer the following empirical formula questions. Show all work! Include units!

1. Dibutyl succinate is an insect repellent used against household ants and roaches. Its composition is 62.58% C, 9.83% H and 27.79% O. Find its empirical formula. $[C_{16}H_{22}O_4]$

$$C: = 5.210658 \text{ mol} + 1.736875 = 3.0000 \quad = 3 \quad \times 2 = 6$$

$$H: = 9.534653 \text{ mol} + 1.736875 = 5.4895 \quad = 5.5 \quad \times 2 = 11$$

$$O: = 1.736875 \text{ mol} + 1.736875 = 1 \quad = 1 \quad \times 2 = 2$$



2. The compound adrenaline contains 56.7% C, 6.56% H, 8.28% N, and 28.37% O by mass. What is the empirical formula for adrenaline?

$$C: = 4.721066 \text{ mol} + 0.591006 = 7.9882 \quad = 8$$

$$H: = 6.495050 \text{ mol} + 0.591006 = 10.9698 \quad = 11$$

$$N: = 0.591006 \text{ mol} + 0.591006 = 1 \quad = 1$$

$$O: = 1.773125 \text{ mol} + 0.591006 = 3.0002 \quad = 3$$



3. A sample of urea contains 0.480 g of C, 0.161 g of H, 1.121 g of N, and 0.640 g of O. What is the empirical formula of urea?

$$C: = 0.039967 \text{ mol} + 0.039967 = 1 \quad = 1$$

$$H: = 0.159406 \text{ mol} + 0.039967 = 3.9664 \quad = 4$$

$$N: = 0.080014 \text{ mol} + 0.039967 = 2.0002 \quad = 2$$

$$O: = 0.040000 \text{ mol} + 0.039967 = 1.0008 \quad = 1$$



Empirical Formula Key Answers

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M. Heyworth,2007 *Forensic Science UGC NET Question Bank Chapterwise Assistant Professor and Lecturer Exams* Mocktime Publication,1988 Forensic Science UGC NET Question Bank Chapterwise Assistant Professor and Lecturer Exams

Chemistry: The Central Science Theodore L. Brown,H. Eugene LeMay Jr.,Bruce E. Bursten,Catherine Murphy,Patrick Woodward,Steven Langford,Dalius Sagatys,Adrian George,2013-10-04 If you think you know the Brown LeMay Bursten Chemistry text think again In response to market request we have created the third Australian edition of the US bestseller Chemistry The Central Science An extensive revision has taken this text to new heights Triple checked for scientific accuracy and consistency this edition is a more seamless and cohesive product yet retains the clarity innovative pedagogy functional problem solving and visuals of the previous version All artwork and images are now consistent in quality across the entire text And with a more traditional and logical organisation of the Organic Chemistry content this comprehensive text is the source of all the information and practice problems students are likely to need for conceptual understanding development of problem solving skills reference and test preparation **Chemistry in a Month** John Wilkinson,2003 Proceedings of 2013 Chinese Intelligent Automation Conference Zengqi Sun,Zhidong Deng,2013-06-28 Proceedings of the 2013 Chinese Intelligent Automation Conference presents selected research papers from the CIAC 13 held in Yangzhou China The topics include e g adaptive control fuzzy control neural network based control knowledge based control hybrid intelligent control learning control evolutionary mechanism based control multi sensor integration failure diagnosis and reconfigurable control Engineers and researchers from academia industry and government can gain an inside view of new solutions combining ideas from multiple disciplines in the field of intelligent automation Zengqi Sun and Zhidong Deng are professors at the Department of Computer Science Tsinghua University China **General Chemistry** Donald A. McQuarrie,Stanley Gill,2011-06-15 This Fourth Edition of McQuarrie s classic text offers a thorough revision and a quantum leap forward from the previous edition Taking an atoms first approach it promises to be another ground breaking text in the tradition of

McQuarrie's many previous works. This outstanding new text, available in a soft cover edition, offers professors a fresh choice and outstanding value. **33 Years NEET Chapterwise & Topicwise Solved Papers CHEMISTRY (2020 - 1988) 15th Edition** Disha Experts, *Basics for Chemistry* David A. Ucko, 2013-09-24. Basics of Chemistry provides the tools needed in the study of General Chemistry such as problem solving skills, calculation methods, and the language and basic concepts of chemistry. The book is designed to meet the specific needs of underprepared students. Concepts are presented only as they are needed and developed from the simple to the complex. The text is divided into 18 chapters, each covering some particular aspect of chemistry such as matter, energy, and measurement; the properties of atoms; description of chemical bonding; study of chemical change; and nuclear and organic chemistry. Undergraduate students will find the book a very valuable academic material.

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