

## ACIDS AND BASES QUIZ

1. Which of the following indicates an acidic solution?

- A) pH = 7.5                      B) pH = 7                      C) pH = 6.5  
D) more than one correct response      E) no correct response

2. Which of the following equations represents an acid-base neutralization reaction?

- A)  $\text{H}_2\text{SO}_4 + \text{Zn} \rightarrow \text{ZnSO}_4 + \text{H}_2$                       B)  $\text{HNO}_3 + \text{KOH} \rightarrow \text{KNO}_3 + \text{H}_2\text{O}$   
C)  $\text{Ba}(\text{OH})_2 + \text{Na}_2\text{SO}_4 \rightarrow \text{BaSO}_4 + 2\text{NaOH}$       D) more than one correct response  
E) none of the above

3. In which of the following pairs of substances are both members of the pair salts?

- A) NaCl and NaOH                      B)  $\text{KNO}_3$  and  $\text{K}_2\text{SO}_4$                       C)  $\text{H}_3\text{PO}_4$  and  $(\text{NH}_4)_3\text{PO}_4$   
D) more than one correct response                      E) none of the above

4. If the pH of a solution increases from 4.0 to 6.0, the hydronium ion concentration

- A) increases by a factor of 1.5.                      B) increases by a factor of 100.  
C) decreases by a factor of 1.5.                      D) decreases by a factor of 100.

5. Which of the following statements concerning Arrhenius acids and Arrhenius bases is incorrect?

- A) In the pure state, Arrhenius acids are covalent compounds.  
B) In the pure state, Arrhenius bases are ionic compounds.  
C) Dissociation is the process by which Arrhenius acids produce  $\text{H}^+$  ions in solution.  
D) Arrhenius bases are also called oxides.

6. Highly basic solutions are:

- A) safe to handle                      B) sweet tasting  
C) corrosive and reactive like highly acidic solutions                      D) corrosive but not reactive

7. When dissolved in water, sodium hydroxide (NaOH):

- A) turns phenolphthalein colourless.      B) produces a high concentration of hydronium ions.  
C) releases  $\text{O}^{2-}$  ions.                      D) produces a high concentration of hydroxide ions.  
E) reacts with carbonates to produce a salt and carbon dioxide gas.

8. Bases taste:

- A) Salty                      B) Bitter                      C) Sweet                      D) Sour

9. The characteristic ion of acids is:

- A) carbonate ion  $\text{CO}_3^{2-}$       B) phosphate ion  $\text{PO}_4^{3-}$       C) hydrogen ion ( $\text{H}^+$ )      D) hydroxide ion ( $\text{OH}^-$ )

10. Which of the following pH values indicates the strongest base?

- A) 4                      B) 7                      C) 11                      D) 1

11. A positive test for a base occurs when:

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