CHEMICAL EQUILIBRIUM MULTIPLE CHOICE QUESTIONS

Read the questions properly before answering.

1. It is a condition that occurs when opposing reactions are proceeding at equal rates.
   a. Chemical kinetics
   b. Chemical equilibriam
   c. Chemical reaction
   d. Chemical equation
2. Which of the following is not included when writing equilibriam constant expression?
   a. gases
   b. aqueous solutions
   c. liquids
   d. pure liquids and solids
3. Equilibrium constant greater than 1 means
   a. equilibrium constant lies to the right , products predominate.
   b. equilibrium constant lies to the left, reactants predominate.
   c. equilibrium constant lies to the right, reactants predominate.
   d. equilibrium constant lies to the left, products predominate.
4. What does it mean when the Keq < 1?
   a. Equilibriam constant lies to the right , products predominate.
   b. Equilibrium constant lies to the left, reactants predominate.
   c. Equilibriam constant lies to the right, reactants predominate.
   d. Equilibrium constant lies to the left, products predominate.
5. What is the equilibrium constant expression of aA + bB ⇌ cC + dD?

   a. $Keq = \frac{[a]^A [b]^B}{[c]^C [d]^D}$

   b. $Keq = \frac{[A]^a [B]^b}{[C]^c [D]^d}$

   c. $Keq = \frac{[C]^c [D]^d}{[a]^A [b]^B}$

   d. $Keq = \frac{[C]^c [D]^d}{[A]^a [B]^b}$

6. Which of the following is not a factor that affects chemical equilibrium?
   a. concentration
   b. temperature
   c. pressure and volume
   d. density
7. Equilibria in which all substances are in the same phases are called
   a. homogeneous equilibria.
   b. heterogeneous equilibria.
   c. equilibrium constant
   d. equilibrium constant expression.
8. Equilibria in which two or more phases are present in the reaction are called
   a. homogeneous equilibria.
   b. heterogeneous equilibria.
   c. equilibrium constant
   d. equilibrium constant expression.

Multiple Choice Concept Questions On Equilibrium

JE Gale

Multiple Choice Concept Questions On Equilibrium:

Unveiling the Magic of Words: A Review of "**Multiple Choice Concept Questions On Equilibrium**"

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