

## Gas Stoichiometry: Multiple Choice

1. What volume of pure oxygen gas,  $O_2$ , measured at 546 K and 101.325 kPa is formed by the complete decomposition of 0.200 mol  $Ag_2O$  according to the following equation:



- A) 1.12 L      B) 4.48 L      C) 2.24 L      D) 8.96 L      E) 17.82 L

ANS: B

1. How many moles of  $KClO_3$  are needed to form 2.8 L of  $O_2$ , measured at STP, according to the following reaction:  $2KClO_3 \rightarrow 2KCl + 3O_2$

- A) 0.083 mol    B) 0.25 mol    C) 0.17 mol    D) 0.50 mol    E) 0.33 mol

ANS: A

1. What volume of chlorine gas, measured at 0°C and 101.325 kPa, is required to react completely with 5.61 g of KOH in the following reaction:  $Cl_{2(g)} + 2KOH \rightarrow KCl + KClO + H_{2O}$

- A) 0.56 L      B) 4.48 L      C) 1.12 L      D) 22.4 L      E) 2.24 L

ANS: C

1. Sodium hydride reacts with excess water to produce sodium hydroxide (aq) and hydrogen gas:  $NaH(s) + H_2O(l) \rightarrow NaOH(aq) + H_2(g)$

A sample of  $NaH$  weighing \_\_\_\_\_ g will produce 982 mL of gas at 28.0 °C and 765 torr, when the hydrogen is collected over water. The vapor pressure of water at this temperature is 28 torr.

- A) 2.93      B) 0.960      C) 0.925      D) 0.0388      E) 925

ANS: C

1. At STP, how many liters of oxygen are required to react completely with 3.6 liters of hydrogen to form water?  $2H_2(g) + O_2(g) \rightarrow 2H_2O(g)$

- A) 1.8 L      B) 3.6 L      C) 2.0 L      D) 2.4 L

ANS: A

1. The molar volume of a gas at STP is \_\_\_\_\_ L.

- A) 0.08206    B) 62.36    C) 1.00    D) 22.4    E) 14.7

ANS: D

1. If 0.5 L of  $O_2(g)$  reacts with  $H_2$  to produce 1 L of  $H_2O(g)$ , what is the volume of  $H_2O(g)$  obtained from 1 L of  $O_{2(g)}$ ?

- A) 0.5 L      B) 1.5 L      C) 2 L      D) 2.5 L

ANS: C

# Multiple Choice Questions On Chemical Stoichiometry

**Basic Concepts of Chemistry and MCQ  
for NEET**



## **Multiple Choice Questions On Chemical Stoichiometry:**

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with choices lettered A through E

Material Tested	Topic Percentage	Number of Questions
Atomic	however do not review too much at any one time	Concentrate on one problem area at a time by reviewing the questions and explanations and by studying our review until you are confident you completely understand the material
Keep track of your scores	By doing so you will be able to gauge your progress and discover general weaknesses in particular sections	You should carefully study the reviews that cover your areas of difficulty as this will build your skills in those areas

**TEST TAKING TIPS**

Although you may be unfamiliar with standardized tests such as the SAT II Chemistry Subject Test there are many ways to acquaint yourself with this type of examination and help alleviate your test taking anxieties. Become comfortable with the format of the exam. When you are practicing to take the SAT II Chemistry Subject Test simulate the conditions under which you will be taking the actual test. Stay calm and pace yourself. After simulating the test only a couple of times you will boost your chances of doing well and you will be able to sit down for the actual exam with much more confidence. Know the directions and format for each section of the test. Familiarizing yourself with the directions and format of the exam will not only save you time but will also ensure that you are familiar enough with the SAT II Chemistry Subject Test to avoid nervousness and the mistakes caused by being nervous. Do your scratchwork in the margins of the test booklet. You will not be given scrap paper during the exam and you may not perform scratchwork on your answer sheet. Space is provided in your test booklet to do any necessary work or draw diagrams. If you are unsure of an answer guess. However if you do guess guess wisely. Use the process of elimination by going through each answer to a question and ruling out as many of the answer choices as possible. By eliminating three answer choices you give yourself a fifty fifty chance of answering correctly since there will only be two choices left from which to make your guess. Mark your answers in the appropriate spaces on the answer sheet. Fill in the oval that corresponds to your answer darkly completely and neatly. You can change your answer but remember to completely erase your old answer. Any stray lines or unnecessary marks may cause the machine to score your answer incorrectly. When you have finished working on a section you may want to go back and check to make sure your answers correspond to the correct questions. Marking one answer in the wrong space will throw off the rest of your test whether it is graded by machine or by hand. You don't have to answer every question. You are not penalized if you do not answer every question. The only penalty results from answering a question incorrectly. Try to use the guessing strategy but if you are truly stumped by a question remember that you do not have to answer it. Work quickly and steadily. You have a limited amount of time to work on each section so you need to work quickly and steadily. Avoid focusing on one problem for too long. Before the Test Make sure you know where your test center is well in advance of your test day so you do not get lost on the day of the test. On the night before the test gather together the materials you will need the next day. Your admission ticket. Two forms of identification e.g. driver's license student identification card or current alien registration card. Two No. 2 pencils with erasers. Directions to the test center. A watch if

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