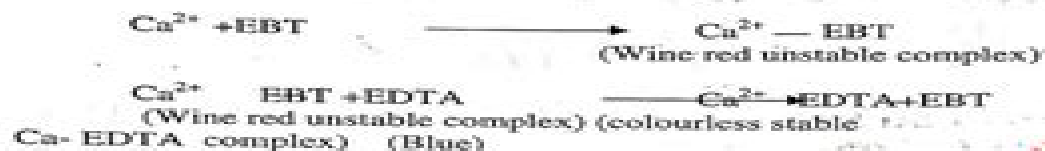
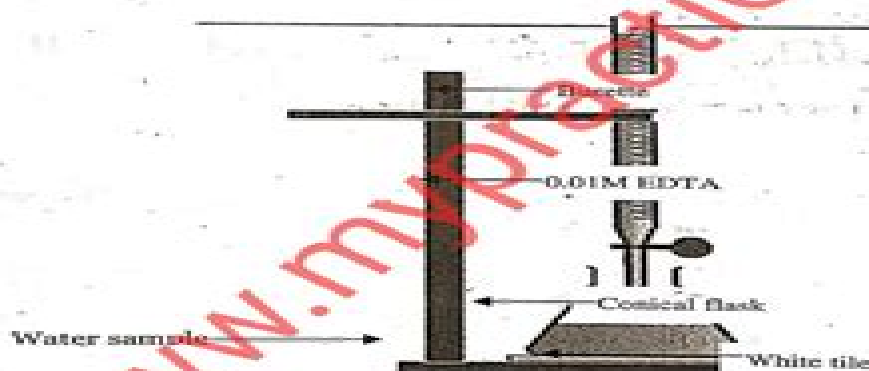


alkaline earth metal ions Ca^{2+} and Mg^{2+} . The EDTA reagent can be used to measure the total quantity of dissolved Ca^{2+} and Mg^{2+} ions in a water sample. Determination of hardness of water is based on the fact that when indicator, Eriochrome Black -T (EBT) is added to hard water in alkaline medium, it forms wine red colour complex with Ca^{2+} and Mg^{2+} ions.



Thus the total hardness of a water sample can be estimated by titration with standard solution of EDTA.

VIII Practical set-up / Circuit diagram / Work Situation



IX Resources Required

S. No.	Name of Resource	Suggested Broad Specification	Quantity
1.	Burette	Borosil glass , Capacity 25 ml /50 ml	One per group
2.	Conical flask	Borosil glass, Capacity 250 ml/100 ml	One per group
3.	Beaker	Borosil glass, Capacity 250 ml/100 ml	One per group
4.	Pipette	Borosil glass, Capacity 25ml / 10 ml,	One per group
5.	EDTA solution	0.01 M	One per group
6.	Buffer solution	pH = 10	As per Requirement
7.	Eriochrome Black T indicator		
8.	Water sample	Hard water sample	

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Table of Contents Msbte Applied Chemistry Answer Key

1. Understanding the eBook Msbte Applied Chemistry Answer Key
 - The Rise of Digital Reading Msbte Applied Chemistry Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Msbte Applied Chemistry Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Msbte Applied Chemistry Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Msbte Applied Chemistry Answer Key
 - Personalized Recommendations
 - Msbte Applied Chemistry Answer Key User Reviews and Ratings
 - Msbte Applied Chemistry Answer Key and Bestseller Lists
5. Accessing Msbte Applied Chemistry Answer Key Free and Paid eBooks
 - Msbte Applied Chemistry Answer Key Public Domain eBooks
 - Msbte Applied Chemistry Answer Key eBook Subscription Services
 - Msbte Applied Chemistry Answer Key Budget-Friendly Options
6. Navigating Msbte Applied Chemistry Answer Key eBook Formats

- ePub, PDF, MOBI, and More
- Msbte Applied Chemistry Answer Key Compatibility with Devices
- Msbte Applied Chemistry Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Msbte Applied Chemistry Answer Key
 - Highlighting and Note-Taking Msbte Applied Chemistry Answer Key
 - Interactive Elements Msbte Applied Chemistry Answer Key
- 8. Staying Engaged with Msbte Applied Chemistry Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Msbte Applied Chemistry Answer Key
- 9. Balancing eBooks and Physical Books Msbte Applied Chemistry Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Msbte Applied Chemistry Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Msbte Applied Chemistry Answer Key
 - Setting Reading Goals Msbte Applied Chemistry Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Msbte Applied Chemistry Answer Key
 - Fact-Checking eBook Content of Msbte Applied Chemistry Answer Key
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
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