

## Intro to Chemical Reactions:

### Test Review Answer Key



1. What do the symbols (s), (l), (g), and (aq) represent? (s) = solid, (l) = liquid, (g) = gas, and (aq) = aqueous
2. Why must chemical equations be balanced? Chemical equations must be balanced to satisfy the Law of Conservation of Mass. The number of atoms of the reactant side must equal the number of atoms on the product side of a chemical equation.
3. How can we be certain that a chemical reaction has taken place? Color change, temperature change, formation of a precipitate, production of gas (bubbles), and light emission.
4. Balance the following chemical equations:
  - a.  $1 \text{ AsCl}_3 + 3 \text{ H}_2\text{O} \rightarrow 3 \text{ HCl} + 1 \text{ As(OH)}_3$
  - b.  $4 \text{ Na} + 1 \text{ O}_2 \rightarrow 2 \text{ Na}_2\text{O}$
  - c.  $1 \text{ Cu} + 2 \text{ H}_2\text{SO}_4 \rightarrow 1 \text{ CuSO}_4 + 1 \text{ SO}_2 + 2 \text{ H}_2\text{O}$
  - d.  $2 \text{ Sb} + 3 \text{ H}_2\text{O} \rightarrow 1 \text{ Sb}_2\text{O}_3 + 3 \text{ H}_2$
  - e.  $1 \text{ La(NO}_3)_3 + 3 \text{ NaOH} \rightarrow 1 \text{ La(OH)}_3 + 3 \text{ NaNO}_3$
  - f.  $1 \text{ Ca(AlO}_2)_2 + 8 \text{ HCl} \rightarrow 2 \text{ AlCl}_3 + 1 \text{ CaCl}_2 + 4 \text{ H}_2\text{O}$
  - g.  $1 \text{ C}_4\text{H}_8 + 6 \text{ O}_2 \rightarrow 4 \text{ CO}_2 + 4 \text{ H}_2\text{O}$
  - h.  $2 \text{ C}_3\text{H}_4\text{O}_3 + 7 \text{ O}_2 \rightarrow 6 \text{ CO}_2 + 8 \text{ H}_2\text{O}$
5. Identify the following types of chemical reactions:
  - a.  $2 \text{ H}_2 + \text{O}_2 \rightarrow 2 \text{ H}_2\text{O}$  **Synthesis or Combustion**
  - b.  $2 \text{ H}_2\text{O} \rightarrow 2 \text{ H}_2 + \text{O}_2$  **Decomposition**
  - c.  $\text{Zn} + \text{H}_2\text{SO}_4 \rightarrow \text{ZnSO}_4 + \text{H}_2$  **Single Replacement**
  - d.  $2 \text{ CO} + \text{O}_2 \rightarrow 2 \text{ CO}_2$  **Synthesis or Combustion**
  - e.  $2 \text{ HgO} \rightarrow 2 \text{ Hg} + \text{O}_2$  **Decomposition**
  - f.  $2 \text{ KBr} + \text{Cl}_2 \rightarrow 2 \text{ KCl} + \text{Br}_2$  **Single Replacement**
  - g.  $\text{CaO} + \text{H}_2\text{O} \rightarrow \text{Ca(OH)}_2$  **Synthesis**
  - h.  $\text{AgNO}_3 + \text{NaCl} \rightarrow \text{AgCl} + \text{NaNO}_3$  **Double Replacement**
  - i.  $2 \text{ H}_2\text{O}_2 \rightarrow 2 \text{ H}_2\text{O} + \text{O}_2$  **Decomposition**
  - j.  $\text{Ca(OH)}_2 + \text{H}_2\text{SO}_4 \rightarrow \text{CaSO}_4 + 2 \text{ H}_2\text{O}$  **Double Replacement**

# Intro To Chemical Reactions Test

**Daniela Niemeyer**



## Intro To Chemical Reactions Test:

**An Introduction to Chemical Theory** Alexander Scott, 1891      *Introduction to the Study of Chemical Reactions* Edmund Drechsel, 1894      Chemical Reaction Hazards Katherine Barton, Richard Rogers, 1997-02-27 This revised edition of a best selling book continues to provide a basis for the identification and evaluation of chemical reaction hazards for chemists engineers plant personnel and students Before undertaking the design of a chemical manufacturing process it is vital that the chemical reactions involved be fully understood potential hazards assessed and safety measures planned Chemical Reaction Hazards aims to help the people responsible for this design and operation to meet the general duties of safety Two major additions to this revised book are the appendices One of these describes 100 incidents illustrating their cause and indicating consequences if appropriate procedures within this guide are not followed The second provides a practical example of a typical chemical reaction hazard assessment from consideration of the process description through experimental testing to the specification of safety measures      *Introduction to Forensic Chemistry* Kelly M. Elkins, 2018-09-03 Chemistry Forensic Science Forensic chemistry is a subdiscipline of forensic science its principles guide the analyses performed in modern forensic laboratories Forensic chemistry's roots lie in medico legal investigation toxicology and microscopy and have since led the development of modern forensic analytic techniques and practices for use in a variety of applications Introduction to Forensic Chemistry is the perfect balance of testing methods and application Unlike other competing books on the market coverage is neither too simplistic nor overly advanced making the book ideal for use in both undergraduate and graduate courses The book introduces chemical tests spectroscopy advanced spectroscopy and chromatography to students The second half of the book addresses applications and methods to analyze and interpret controlled substances trace evidence questioned documents firearms explosives environmental contaminants toxins and other topics The book looks at innovations in the field over time including the latest development of new discernible chemical reactions instrumental tools methods and more Key features Nearly 300 full color figures illustrating key concepts and over 20 case studies Addresses all the essential topics without extraneous or overly advanced coverage Includes full pedagogy of chapter objectives key terms lab problems end of chapter questions and additional readings to emphasize key learning points Includes chemical structures and useful spectra as examples Fulfills the forensic chemistry course requirement in FEPAC accredited programs Includes a chapter on Chemical Biological Radiological Nuclear and Explosive CBRNE materials Comprehensive and accessible without being overly technical Introduction to Forensic Chemistry will be a welcome addition to the field and an ideal text designed for both the student user and professor in mind Course ancillaries including an Instructor's Manual with Test Bank and chapter PowerPoint lecture slides are available with qualified course adoption      **DAT: Dental Admissions Test: Includes 3 Full Length Practice Tests + Online Access to Video Tutorials** Barron's Educational Series, Joseph DiRienzo, John J. Ference, Nicole D. Cornell, Edwin H. Hines, John Swartwood, 2018-05-15 This brand new manual prepares dental school

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Introduction to Forensic Science and Criminalistics, Second Edition Howard A. Harris, Henry C. Lee, 2019-06-20 This Second Edition of the best selling Introduction to Forensic Science and Criminalistics presents the practice of forensic science from a broad viewpoint The book has been developed to serve as an introductory textbook for courses at the undergraduate level for both majors and non majors to provide students with a working understanding of forensic science The Second Edition is fully updated to cover the latest scientific methods of evidence collection evidence analytic techniques and the application of the analysis results to an investigation and use in court This includes coverage of physical evidence evidence collection crime scene processing pattern evidence fingerprint evidence questioned documents DNA and biological evidence drug evidence toolmarks and firearms arson and explosives chemical testing and a new chapter of computer and digital forensic evidence Chapters address crime scene evidence laboratory procedures emergency technologies as well as an adjudication of both criminal and civil cases utilizing the evidence All coverage has been fully updated in all areas that have advanced since the publication of the last edition Features include Progresses from introductory concepts of the legal system and crime scene concepts to DNA forensic biology chemistry and laboratory principles Introduces students to the scientific method and the application of it to the analysis to various types and classifications of forensic evidence The authors 90 plus years of real world police investigative and forensic science laboratory experience is brought to bear on the application of forensic science to the investigation and prosecution of cases Addresses the latest developments and advances in forensic sciences particularly in evidence collection Offers a full complement of instructor s resources to qualifying professors Includes full pedagogy including learning objectives key terms end of chapter questions and boxed case examples to encourage classroom learning and retention Introduction to Forensic Science and Criminalistics Second Edition will serve as an invaluable resource for students in their quest to understand the

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