

# Intro to Acids and Bases

Name: \_\_\_\_\_  
 Class: \_\_\_\_\_  
 Date: \_\_\_\_\_

## Define each term: Definitions of acids and bases:

### Arrhenius-definitions:

- acid substances that release dissolved cations hydrogen ( $H^+$ ) (actual  $H^+$  cations bonded to water as the hydronium ion,  $H_3O^+$ , and  $H^+$  (actual  $H_2O^+$ ) and hypothetical equivalents)
- base substances which release dissolved water hydroxide ( $OH^-$ )

### Brønsted-Lowry definitions:

- acid substances that donate a proton ( $H^+$ ) (also accepted)
- base substances which accept a proton ( $H^+$ ) (also donated)

Table 1: Representative reactions for  $H^+$  (also known as the Lewis acid) and bases according to Arrhenius:

Reaction	Eq.
1. $HCl(aq) + H_2O(l) \rightleftharpoons H_3O^+(aq) + Cl^-(aq)$	$H^+(aq) + H_2O(l)$
2. $HNO_3 + H_2O \rightleftharpoons H_3O^+ + NO_3^-$ (actual $H^+$ is $H_3O^+$ )	$H^+ + H_2O(l)$
3. $HNO_3(aq) + H_2O(l) \rightleftharpoons H_3O^+(aq) + NO_3^-(aq)$	$H^+(aq) + H_2O(l)$
4. $H_2SO_4 + H_2O(l) \rightleftharpoons H_3O^+ + HSO_4^-$	$H^+(aq) + H_2O(l)$
5. $HCl + H_2O(l) \rightleftharpoons H_3O^+ + Cl^-$	$H^+ + H_2O(l)$

## Critical Thinking Questions

1. List all of the components for Table 1 that are Arrhenius acids:

eq. 1:  $HCl(aq)$ ,  $H_3O^+(aq)$ ,  $Cl^-(aq)$

2. List all of the components for Table 1 that are Arrhenius bases:

$OH^-(aq)$

3. List all of the components in Table 1 that are Brønsted-Lowry acids:

$HCl(aq)$ ,  $HNO_3(aq)$ ,  $H_2SO_4(aq)$ ,  $H_2SO_4(l)$ ,  $HCl(l)$

4. List all of the components in Table 1 that are Brønsted-Lowry bases:

$OH^-(aq)$ ,  $H_2O(l)$ ,  $Cl^-(aq)$

5. Is it possible for one acid to act as a base? Explain:

Yes, because  $H_2SO_4$ ,  $HNO_3$ ,  $HCl$  can act as bases

6. Do acid substances contain Brønsted-Lowry acids and/or bases? Explain:

yes, because  $HCl$  is  $H^+$  and  $H_2O$  is  $OH^-$  and  $H_2SO_4$  and  $HNO_3$  are acids and  $Cl^-$  and  $NO_3^-$  are bases

7. Is it possible for one acid substance to act as a Brønsted-Lowry acid and/or base? Explain:

# Intro To Acids And Bases Worksheet Answers

## Chemquest 48

**IM Harris**



## **Intro To Acids And Bases Worksheet Answers Chemquest 48:**

This book delves into Intro To Acids And Bases Worksheet Answers Chemquest 48. Intro To Acids And Bases Worksheet Answers Chemquest 48 is an essential topic that needs to be grasped by everyone, ranging from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Intro To Acids And Bases Worksheet Answers Chemquest 48, encompassing both the fundamentals and more intricate discussions.

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