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# Grade Mathematics Paperjune Exam

**AW Chickering**



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web solution a titrant volume 0 mL the solution  $pH$  is due to the acid ionization of  $HCl$  because this is a strong acid the ionization is complete and the hydronium ion molarity is 0.100 M the  $pH$  of the solution is then  $pH = -\log 0.100 = 1.000$   $pH = 0.100 = 1.000$  b titrant volume 12.50 mL

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solutions of weak acids with the indicated  $pK_a$  values

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web ab acid base titration curves purpose the purpose of this lab is to determine the shape of acid base titration curves theoretically and experimentally and to use an acid base titration to measure the effectiveness of antacid tablets summary a titration curve is the plot of the  $pH$  value of a solution vs the amount of titrant added

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Volume of $NaOH$ added ( $mL$ )	$pH$ of the solution
4	6
4	6
5	9
5	9

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