

Determination of concentration of a Dextrose solution of unknown strength using an Ostwald Viscometer.

Aim: To determine the concentration of a Dextrose solution of unknown strength using an Ostwald Viscometer and to study the variation of this solution's viscosity as a function of temperature.

Apparatus Required: 1. Ostwald Viscometer, Thermometer
2. Water, Ice water, glass jar
3. Hot plate, standard volumetric flask 25 ml

Chemicals Required: 1. Dextrose, Distilled water

Theory: Viscosity: The internal friction of a liquid which resist the flow of it is known as viscosity.

The force it implies can be measured in terms of the tangent force which is needed to keep the speed of layers constant, this force is called viscous force.

$$\eta = \frac{\pi \mu^4 P L}{8 \pi V L}$$

$$\text{Volume rate law} = \frac{V}{t} = \frac{P \pi \mu^4}{8 \eta L}$$

V = Volume of the liquid

t = time of flow of liquid

η = Viscosity coefficient

P = Hydrostatic pressure

L = Distance travelled by the liquid during time t .

The units of viscosity are called Poise (P). The official SI unit for absolute viscosity is kg/ms or (Pascal second, Pa.s).

Ostwald viscometer is a commonly used viscometer, which consists of a U-shaped glass tube held vertically. For more accurate

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Msc Organic Chemistry Practical Manual

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Table of Contents Msc Organic Chemistry Practical Manual

1. Understanding the eBook Msc Organic Chemistry Practical Manual
 - The Rise of Digital Reading Msc Organic Chemistry Practical Manual
 - Advantages of eBooks Over Traditional Books
2. Identifying Msc Organic Chemistry Practical Manual
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Msc Organic Chemistry Practical Manual
 - User-Friendly Interface
4. Exploring eBook Recommendations from Msc Organic Chemistry Practical Manual
 - Personalized Recommendations
 - Msc Organic Chemistry Practical Manual User Reviews and Ratings
 - Msc Organic Chemistry Practical Manual and Bestseller Lists
5. Accessing Msc Organic Chemistry Practical Manual Free and Paid eBooks
 - Msc Organic Chemistry Practical Manual Public Domain eBooks
 - Msc Organic Chemistry Practical Manual eBook Subscription Services
 - Msc Organic Chemistry Practical Manual Budget-Friendly Options

6. Navigating Msc Organic Chemistry Practical Manual eBook Formats
 - ePub, PDF, MOBI, and More
 - Msc Organic Chemistry Practical Manual Compatibility with Devices
 - Msc Organic Chemistry Practical Manual Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Msc Organic Chemistry Practical Manual
 - Highlighting and Note-Taking Msc Organic Chemistry Practical Manual
 - Interactive Elements Msc Organic Chemistry Practical Manual
8. Staying Engaged with Msc Organic Chemistry Practical Manual
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Msc Organic Chemistry Practical Manual
9. Balancing eBooks and Physical Books Msc Organic Chemistry Practical Manual
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Msc Organic Chemistry Practical Manual
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Msc Organic Chemistry Practical Manual
 - Setting Reading Goals Msc Organic Chemistry Practical Manual
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Msc Organic Chemistry Practical Manual
 - Fact-Checking eBook Content of Msc Organic Chemistry Practical Manual
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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