

### Introduction

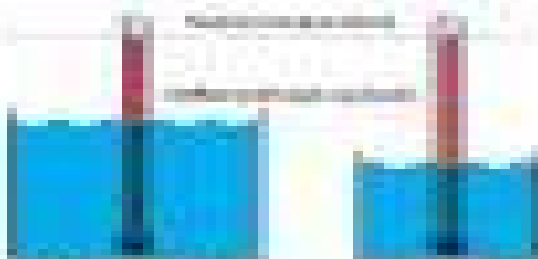
Many natural and man-made systems behave according to the principles of thermodynamics (the science). Examples of this are listed below:

Thermodynamics is the study of the \_\_\_\_\_, the distribution of \_\_\_\_\_ that can affect these systems and the surroundings (all included thermodynamically systems).

### Thermodynamics and statistics

- If you add any additional variables to thermodynamics, the world is not really closed.
- Heat = 0
- Temperature = 0
- Entropy = 0

Heat	Temperature
<ul style="list-style-type: none"> <li>• Heat is the amount of energy that is added to a system.</li> <li>• e.g. you can add heat to a gas, fluid or a solid material.</li> <li>• Heat is the energy that is added to a system.</li> <li>• Heat is the energy that is added to a system.</li> </ul>	<ul style="list-style-type: none"> <li>• Temperature is the average energy that is "per unit" mass of a substance.</li> </ul>



<p>Thermodynamic system is a system that is defined by its boundaries.</p> <p>Thermodynamic system is a system that is defined by its boundaries.</p> <p>Thermodynamic system is a system that is defined by its boundaries.</p>
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### Thermodynamics

- Heat transfer to the system
- Heat transfer from the system
- Heat transfer to the system
- Heat transfer from the system



# Intro To Thermochemistry Activities

**Y Pai**



## **Intro To Thermochemistry Activities:**

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