



Index no:

Science

Time: 2.00 Hours

Part – I

➤ **Underline the correct answer**

- Identify the auto trophic micro organism
 1) Virus 2) Fungus 3) Algae 4) Protozoa
- A disease that spread due to bacteria
 1) Malaria 2) Tuberculosis 3) Rabbits 4) Ebola
- A part of eye that is focused on image of an object?
 1) Ciliary muscle 2) Iris 3) Eye lens 4) Retina
- A part of ear that contributes the body balance
 1) Cochlea 2) Tympanic membrane
 4) Ossicles 4) Semi circular canals
- Identify the homo atomic molecule
 1) CH_4 2) H_2O 3) O_2 4) NH_3
- Select the homogeneous mixture
 1) Muddy water 2) Cement mixture
 3) Fruit mixture 4) Sea water
- A valve found between left atrium and left ventricle
 1) Bicupid valve 2) Tricuspid valve
 3) Semi lunar valve 4) Pulmonary valve
- Standard unit to measure the area
 1) M^2 2) N 3) Nm^{-2} 4) kg
- An instance that increase the pressure by reducing area
 1) Broader tyre 2) Use wooden plank to walk on muddy
 4) Tip of iron nail 4) Broader band of a school bag
- Which method used to separate the components of crude oil
 1) Crystallization 2) Evaporation
 3) Steam distillation 4) Fractional distillation

Grade 9 Science Past Papers

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the boiling point of an organic liquid is the temperature at which its vapour pressure equals the atmospheric pressure over the liquid or it is the temperature at which the vapour and liquid phases are in equilibrium at agiven pressure the boiling point is considered as a criterion of purity of a compound and is useful for identification of

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given that the boiling points of ethanol and methanol 78 c and 64 c respectively are far off in a significant way this seems like a valid conclusion conclusion thermometer calibrations are necessary in order to collect reliable information as precise calculations are imperative to gathering correct results

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melting point 1 a push the open end of a capillary tube into the powdered acetamide b move the powder to the closed end of the capillary tube by tapping it on the table repeat until the the powdered acetamide occupies 1 2 mm of the capillary tube end c

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the percentage yield for benzoic acid and acetanilide acid being high indicated the initial sample had a low concentration of impurities with the melting point being in a specific

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determination of boiling points introduction the boiling point of a compound is the temperature at which it changes from a liquid to a gas this is a physical property often used to identify substances or to check the purity of the compound it

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clean up solid and liquid spills immediately true g conclusions after conducting the boiling point determination lab it can be concluded that the unknown substance 193 is isopropyl alcohol on account of its observed boiling point of 88 c

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conclusion after completing this experiment the melting point of three different unknown compounds was obtained using the mel temp apparatus compound a had a melting point range of 114 6 115 5 c thus revealing it to be acetanilide compound b had a melting point range of 122 6 124 5 c thus

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laboratory partners samantha and mackenzie goals □ correctly determine the boiling point of an unknown liquid □ measure the refractive index of an unknown □ use the gas chromatogram to determine the mass percentage of a mixture and

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50 c so in order to convert an experimental measurement taken in calgary which is at higher altitude than sea level and so is at a lower pressure to that reported for sea level higher pressure one needs to add a correction factor since the boiling point at sea level is higher than that at higher altitudes

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introduction in chemistry one may encounter an unknown substance that needs to be determined by physical properties such as boiling point melting point density solubility etc in this experiment i am to figure out the boiling point of a mystery substance in order to identify it chemical responsibility this experiment requires the handling of a bunsen burner boiling

[6 2b step by step procedures for boiling point determination](#) - Mar 15 2023

there are a variety of methods by which a sample's boiling point can be determined including distillation reflux and by using a thiele tube the most straightforward method uses a thiele tube and has the advantage of using less than 0.5 ml of material