

Name : _____ Date : _____

Acids and Bases Strength

Write True (T) or False (F).

- i. Stronger acids have weaker conjugate bases and weaker acids have stronger conjugate bases. _____
- ii. Stronger bases have stronger conjugate acids and weaker bases have weaker conjugate acids. _____

Write equations that show how the following reactions occur in water.

- i. F^- acts as a base. _____
- ii. HNO_2 acts as an acid. _____
- iii. $\text{Fe}(\text{H}_2\text{O})_6^{3+}$ acts as an acid. _____
- iv. HCO_3^- acts as a base. _____
- v. HCO_3^- acts as an acid. _____
- vi. $\text{Al}(\text{H}_2\text{O})_5(\text{OH})^{2+}$ acts as a base. _____

What volume of 0.45 M LiOH would be needed to neutralize 60.0 mL of 0.15 M HI?

Calculate the pH values of the following solutions of strong acids and bases.

- i. 0.001 M HCl
- ii. 0.76 M KOH
- iii. 2.8×10^{-4} M $\text{Ba}(\text{OH})_2$

Lecture 29 Acids And Bases I Worksheet

L Cohen



Lecture 29 Acids And Bases I Worksheet:

Introduction to Chemistry, Laboratory Manual T. R. Dickson, 1994-12-23 Teaches chemistry by offering a dynamic provocative and relevant view of the topic and its importance to society and our daily lives Three themes are stressed throughout the text developing chemical thinking and a chemical vision learning problem solving methods and utilizing group work and discussion activities These themes involve and engage the students in their own learning processes they are challenged to be active The presentation of topics has been altered to include a new chapter which introduces the students to scientific thinking and shows that chemistry involves interesting and relevant topics The reorganization presents many core concepts in the first five chapters preparing students for later chapters In addition the author has added vignettes throughout the chapters referring to health technology the environment and society as well as to specific tools of direct use to students *Current Index to Journals in Education* ,1977

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