



Fisher Scientific

Fisher Isotemp® Low Temperature Incubator Model 307C Catalog Number 11-679-25C

Fisher Low Temperature Incubator offers laboratories precise temperature control over the -10°C to 60°C range. Performance meets requirements for the preservation of vaccines, biologicals, incubation of bacterial cultures, and determination of bio-chemical oxygen demand of sewage. The extended operating temperature range also includes temperatures associated with drug stability projects, dairy product evaluation and entomological studies. The spacious chamber accommodates up to 300 standard BOD bottles, or similar containers. The 5 reinforced shelves can handle up to 60 lbs of sample weight each.

Features include:

- 5 Removable shelves
- 1 Removable storage basket
- 6 Molded door shelves
- Door Lock with key
- Microprocessor based temperature controller, with temperature readout to .1°C
- 3 Cooling control modes: High precision with cooling, High precision without cooling and Frost free with variable cooling
- Solid state relay for heater
- Over/ under temperature safety relay and alarm Over temperature thermostat
- Compressor relay for energy conservation
- Compressor overload relay
- RTD temperature probe
- Protected setpoint mode to avoid accidental change
- Convenience outlet inside chamber, 2 amp
- Interior light with door switch

WARNING

- This manual must be carefully read and thoroughly understood before operating the unit, failure to follow directions or precautionary measures could result in serious adverse effects.
- This equipment must be used only as specified in these instructions.
- This equipment is intended for indoor use only.
- This equipment must be earth grounded for safe operation.
- Maximum ratings of accessories that may be plugged into convenience outlet: Load current 2 Amps. @ 115VAC, Leakage Current 250 micro Amps.

CAUTION

The aluminum evaporator and other portions of these units should not be exposed to the corrosive effects of acidic or caustic materials. Extreme care must be exercised if such materials are stored within to prevent voiding the warranty. This incubator is NOT suitable for flammable material storage.

SPECIFICATIONS

Power Requirements:

115V, 8.9A, 60 Hz (Including 2 A for convenience outlet)

Dimensions:

Height:	70 in (179cm)
Width:	32 in (81cm)
Depth:	28.5 in (72cm)
Volume:	20.3 cu ft (0.58m ³)
Shelf Data:	5 in chamber, 6 on door, 1 basket
Capacity:	300 BOD bottles

Shipping Weight:

283 lb. (128 kg)

Operating Environment:

Temperature Range:	10°C to 40°C
Humidity Range:	0% to 90% RH
Maximum Altitude:	6600 ft (2 km)
Air clearances:	Back, top and sides 3.0" minimum
Overvoltage Category	II (IEC664)
Pollution Degree 2 (IEC664)	

Performance Characteristics:

Operating Range:	-10° to +60°C
Uniformity:	±0.5°C ¹
Stability:	±0.2°C ¹
Display Readability:	0.1°C

¹ Per ASTM method E 1292-94, cooling modes 00 and 100

PACKING LIST

Qty	Item
1	Low Temperature Incubator
1	Instructions

UNPACKING and INSPECTION

NOTE: If damage is observed, keep the shipment intact (including the carton and all packing material) and file a claim with the final carrier.

INSTALLATION

The Low Temperature Incubator should be installed on a level, stable floor. Use the adjustable feet to level the incubator. Locate the unit so that there is at least 3 inches of clearance around the sides, top and back to ensure sufficient ventilation. Note: When cooling is enabled, heat is dissipated on sides of unit resulting in side temperatures of 35 to 55°C. To prevent heat build up within the unit, do not install it where it will be exposed to strong sunlight, nor near radiators, furnaces, or other sources of heat. Do not install the incubator where flammable or corrosive atmospheres may exist. The Low Temperature Incubator is not to be used out-of-doors or where excessive moisture or contamination could cause degrading of mechanical or electrical components.

1. Unbolt and remove the wooden slats underneath unit.
2. Remove all protective tape from doors and shelves.
3. Clean interior and exterior with warm soap and water. Shipping tape residue can be removed using isopropyl alcohol.
4. Uncoil line cord from rear of unit. Connect the plug to a power outlet that complies with the electrical requirements specified on the unit's label and with proper safety ground connection. If line cord replacement is necessary, use CSA types: SJ, SJO or SJT, 16 GA, 3 cond. or equivalent.

Fisher Scientific Model 525f Manual

Michael Seilmaier



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a net ionic equation is the most accurate representation of the actual chemical process that occurs writing these equations requires a familiarity with solubility rules acid base reactivity

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apr 25 2017 this is understood when you consider the two chemicals appear unchanged before and after the reaction therefore the net ionic equation is $2\text{Cl}^- + 2\text{Ag}^+ \rightarrow 2\text{AgCl}$

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ions that remain essentially unchanged during a reaction are called spectator ions this means you can ignore them when you write the ionic equation you only need to model how the solid

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may 11 2017 in the net ionic equation any ions that do not participate in the reaction called spectator ions are excluded as a result the net ionic equation shows only the species that

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oct 2 2019 the net ionic equation is a chemical equation for a reaction that lists only those species participating in the reaction the net ionic equation is commonly used in acid base

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the first step to writing a net ionic equation is balancing the chemical equation present let s use the reaction between sodium chloride and silver nitrate as an example next we write the

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highlights learning objectives by the end of this section you will be able to derive chemical equations from narrative descriptions of chemical reactions write and balance chemical

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the net ionic equation is the chemical equation that shows only those elements compounds and ions that are directly involved in the chemical reaction notice that in writing the net ionic

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what remains when the spectator ions are removed is called the net ionic equation which represents the actual chemical change occurring between the ionic compounds cl aq

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this net ionic equation indicates that solid silver chloride may be produced from dissolved chloride and silver i ions regardless of the source of these ions these molecular and

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write the overall chemical equation the complete ionic equation and the net ionic equation for the reaction of aqueous silver fluoride with aqueous sodium phosphate to give solid silver

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the complete ionic equation for this reaction is as follows equation 4 13 2ag aq 2f aq 2nh₄ aq cr₂o₇²⁻ aq ag₂cr₂o₇ s 2nh₄ aq 2f aq because two nh₄ aq

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