

Unit 1 • Observations, Models, & Experiments**SCIENTIFIC NOTATION**

Change the following to Scientific Notation (maintain the number of significant figures):

1. 5.280 = 5.280

11. 2,560 = 2.56 × 10³

2. 2,000 = 2 × 10³

12. .0009 = 9 × 10⁻⁴

3. 15 = 1.5 × 10¹

13. 8,900,000 = 8.9 × 10⁶

4. 6,589,000 = 6.589 × 10⁶

14. .0920 = 9.20 × 10⁻²

5. 70,400,000,000 = 7.04 × 10¹⁰

15. 6,300 = 6.3 × 10³

6. .00263 = 2.63 × 10⁻³

16. .90 = 9.0 × 10⁻¹

7. .00589 = 5.89 × 10⁻³

17. 250 = 2.5 × 10²

8. .006 = 6 × 10⁻³

18. .006087 = 6.087 × 10⁻³

9. .400 = 4.00 × 10⁻¹

19. 500,000 = 5 × 10⁵

10. .08060 = 8.060 × 10⁻²

20. .0000000105 = 1.05 × 10⁻⁸

Make the following Metric System conversions using "unit analysis" (you may use scientific notation):

1. 100 mg $\times \frac{1 \text{ g}}{1000 \text{ mg}}$ = .1 g

2. 20 cm $\times \frac{1 \text{ m}}{100 \text{ cm}}$ = .2 m

3. 50 L $\times \frac{1 \text{ kL}}{1000 \text{ L}}$ = .05 kL

4. 22 g $\times \frac{100 \text{ cg}}{1 \text{ g}}$ = 2200 cg

5. 825 cm $\times \frac{1 \text{ m}}{100 \text{ cm}} \times \frac{1 \text{ km}}{1000 \text{ m}}$ = 8.25 × 10⁻³ km

6. 2,350 kg $\times \frac{1000 \text{ g}}{1 \text{ kg}}$ = 2.35 × 10⁶ g

7. 19 mL $\times \frac{1 \text{ L}}{1000 \text{ mL}} \times \frac{100 \text{ cL}}{1 \text{ L}}$ = 1.9 cL

8. 52 km $\times \frac{1000 \text{ m}}{1 \text{ km}}$ = 52000 m

9. 36 m $\times \frac{100 \text{ cm}}{1 \text{ m}}$ = 3600 cm

10. 18 cm $\times \frac{1 \text{ m}}{100 \text{ cm}} \times \frac{1000 \text{ mm}}{1 \text{ m}}$ = 180 mm

11. 6 g $\times \frac{1000 \text{ mg}}{1 \text{ g}}$ = 6000 mg

12. 4,259 mg $\times \frac{1 \text{ g}}{1000 \text{ mg}}$ = 4.259 g

Estimating Length Using Scientific Notation Answer Key

CO Houle



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