Specific Heat Worksheet

$C_p = q/m\Delta T$, where q = heat energy, m = mass, and T = temperature

A 15.75-g piece of iron absorbs 1086.75 joules of heat energy, and its temperature changes

from 25°C to 175°C. Calculate the heat capacity of iron (Cp = .46 J/g*c) 1086.75 = 15.75g .150 . Cp (Cp = .46 J/g*c) 1086.75 = 2362.5 Cp

How many joules of heat are needed to raise the temperature of 10.0 g of aluminum from 22°C to 55°C, if the specific heat of aluminum is 0.90 J/g°C?

Q = Cp · m · AT -90(10)(33) = 297 J \ 33

 To what temperature will a 50.0 g piece of glass raise if it absorbs 5275 joules of heat and its heat capacity is 0.50 J/g°C? The initial temperature of the glass is 20.0°C.

5275 = .50(50)AT Tr = 211 2319C

Calculate the heat capacity of a piece of wood if 1500.0 g of the wood absorbs 6.75×10⁴ joules
of heat, and its temperature changes from 32°C to 57°C.

67500 = Cp. 1500 · 25°C [1.8 J/gc]

 100.0 mL of 4°C water is heated until its temperature is 37°C. If the specific heat of water is 4.18 J/g°C, calculate the amount of heat needed to cause this rise in temperature.

Q = 100g · 4.18 · 33°C [13794]

25.0 g of mercury is heated from 25°C to 155°C, and absorbs 455 joules of heat in the process. Calculate the heat capacity of mercury.

455 = Cp · 25g · 130 (-14 = Cp)

 What is the Specific heat capacity of silver metal if 55.00 g of the metal absorbs 198 joules of heat and the temperature rises 15.0°C?

198 = Cp - 55g - 15°C (Cp = -24 J/gc)

 What is the change in temperature of 150.0 g of chloroform if it absorbs 1000.0 joules of heat, and the specific heat of chloroform is 0.96 J/g*C?

1000 = 96 (150g) AT 1000 = 144 A T 694 = AT

6.9°C

Answers.

- Cp = .46 J/g°C
- 2. Q = 297 J
- 3. Innat = 231 °C
- C_P = 1.8 J/Q*C
- 5. Q = 13800 V 13794
- 6. Cp = 14 J/g*C
- 7. Co ≈ 24 J/g°C
- 8. AI = 6.9 °C

<u>Ii Heat Problems Answers</u>

Liying Dong



Ii Heat Problems Answers Book Review: Unveiling the Power of Words

In a world driven by information and connectivity, the energy of words has become more evident than ever. They have the ability to inspire, provoke, and ignite change. Such is the essence of the book **Ii Heat Problems Answers**, a literary masterpiece that delves deep in to the significance of words and their impact on our lives. Published by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we shall explore the book is key themes, examine its writing style, and analyze its overall affect readers.

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