

Ans. ${}_{11}\text{Li}$ - Electronic Configuration (2, 1)
 ${}_{16}\text{S}$ - Electronic Configuration (2, 8, 6)
 ${}_{11}\text{Na}$ - Electronic Configuration (2, 8, 1)
 ${}_{15}\text{P}$ - Electronic Configuration (2, 8, 5)
 Here Silicon (Si), Sodium (Na) and Phosphorus (P) have occupied 3 shells. Hence they belong to period 3.

b. ${}_{11}\text{Li}$, ${}_{21}\text{Sc}$, ${}_{20}\text{Ca}$, ${}_{16}\text{S}$, ${}_{18}\text{Ar}$ which of these elements belong to the second group?

Ans. ${}_{11}\text{Li}$ - Electronic Configuration (1)
 ${}_{21}\text{Sc}$ - Electronic Configuration (2, 8, 3)
 ${}_{20}\text{Ca}$ - Electronic Configuration (2, 8, 8, 2)
 ${}_{16}\text{S}$ - Electronic Configuration (2, 8, 6)

Hsc Solution Of Chemistry Chp 10

C. M. Roebuck



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