

Please remember to put units on your answers, round to the correct number of sig figs, and circle your answer to any problems. Use scrap paper to show work for credit.

- Round your answers to the following to the correct # of sig figs.
 - $9.767 + 3.45 + 0.73 =$
 - $(4.0)(3170) =$
 - $(456.0)/(1.8) =$
- $192.^{\circ}\text{C} =$ _____ k $63.6 \text{ k} =$ _____ $^{\circ}\text{C}$
- The boiling point of liquid helium is -269°C . What is this temperature in kelvin?
- The temperature required to melt NaCl is 801 K. What is this temperature in Celsius?
- Which of the following is the largest volume?
 - $1.3 \square 10^7 \text{ pL}$
 - $1.5 \square 10^8 \text{ nL}$
 - $5.0 \square 10^5 \text{ mL}$
 - $7.0 \square 10^2 \text{ mL}$
 - $3.0 \square 10^{-1} \text{ L}$
- Wirtz, late for work, gets busted on Rt 302 doing 67 miles/hr. Convert this speed to feet/min.
- A rectangular box has dimensions of 20.0 cm x 15.0 cm x 8.00 cm. Calculate the volume of the box in liters.
- If the fuel efficiency of an automobile is 22 miles per gallon, what is its fuel efficiency in kilometers per liter? (1 km = 0.621 mile, 1.000 L = 1.057 quarts, 4 quarts = 1 gallon)

Honors Chemistry Gas Laws Review

**Barron's Educational Series, Mark
Kernion, Joseph A. Mascetta**



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