

*Honors Chemistry Midterm: Review Packet***PART 1**

1. Write in the number of significant figures for each of the following measurements:

3 3.00 mL 3 0.00290 g 4 50.00 m 2 0.070 kg 1 400 L

2. Using the following masses: 0.0562 kg, 124.213 g and 1635 mg.
- ← convert to grams*
-
- The answer should be reported as:
-
- g

56.2g 124.213g 1.635g

3. a) A 5.75 mL sample of mercury has a measured mass of 77.05 g. The density is
- 13.49/mL

$$D = \frac{m}{V} = \frac{77.05g}{5.75mL} =$$

- b) Mercury's accepted density is 13.53 g/mL. The % error in your measurement is:
- .9617%

$$\frac{|\text{actual} - \text{exp}|}{\text{actual}} \times 100 \quad \frac{|13.53g/mL - 13.49g/mL|}{13.53g/mL} \times 100$$

4. In the diagram below:



Is this person accurate? no - not in center

Is this person precise? yes - close together

5. Identify the following as a chemical change (CC), physical change (PC), chemical property (CP), or physical property (PP):

a) burning wood CC b) mass of wood PP
c) flammability of wood CP d) chopping up wood PC

Honors Chemistry Review Packet Answers

Emilie Sanchez



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the recommended strategies that support it are useful for anyone moving on to a more advanced stage of education so this book also has an intended audience of students preparing to go to high school graduate school or professional school In a conversational tone and liberally illustrated by anecdotes of past students the author combines introducing readers to concepts like Bloom s Taxonomy to illuminate the difference between studying and learning fixed and growth mindsets as well as to what brain science has to tell us about rest nutrition and exercise together with such highly specific learning strategies as how to read a textbook manage their time and take tests With engaging exercises and thought provoking reflections this book is an ideal motivational and practical text for study skills and first year experience courses

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