Date

## Chemistry — Unit 5 Worksheet 3 Empirical and Molecular Formulas

Show all your work when solving the following problems. Be sure to include units and label your answer.

 Find the empirical formula of a compound containing 32.0 g of bromine and 4.9 g of magnesium.

32.0 g × <u>1 mole</u> = 0.401 mole Br 79.9 g

1:2 ratio

4.9 g × 1 mole = 0.20 mole Mg 24.31 g

MgBrz

What is the empirical formula of a carbon-oxygen compound, given that a 95.2 g sample of the compound contains 40.8 g of carbon and the rest oxygen?
95.2 g - 40.8 g = 54.4 g O

 $40.8 \text{ g} \times 1 \text{ mole} = 3.40 \text{ moles C}$ 12.0 g

1:1 ratio

54.4 g x× <u>1 mole</u> = 3.40 moles O 16.0g CO

3. A compound was analyzed and found to contain 9.8 g of nitrogen, 0.70 g of hydrogen, and 33.6 g of oxygen. What is the empirical formula of the compound?

9.8 g × 1 mole = 0.70 moles N 14.0g

1:1:3 ratio

0.70 g × <u>1 mole</u> = 0.69 moles H 1.01 g HNO<sub>3</sub>

33.6 g × 1 mole = 2.10 moles O 16.0 g

 A compound composed of hydrogen and oxygen is found to contain 0.59 g of hydrogen and 9.40 g of oxygen. The molar mass of this compound is 34.0 g/mol. Find the empirical and molecular formulas.

0.59 g × 1 mole = 0.58 mole H 1.01 g 9.40 g × 1 mole = 0.588 mole O 1:1 ratio; empirical formula: HO empirical formula mass: 17.0 g/mol molecular formula mass: 34.0 g/mol mfm is 2x efm so the molecular formula is: H-O-

16.0 g

## **Empirical Formula Answer Sheet**

**Frank Brescia** 



#### Reviewing Empirical Formula Answer Sheet: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is truly astonishing. Within the pages of "**Empirical Formula Answer Sheet**," an enthralling opus penned by a highly acclaimed wordsmith, readers attempt an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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