

2. Show that each of the following arguments is valid in *SD*.

$$\begin{array}{l} \text{a. } A \supset \neg B \\ \quad \neg B \supset C \\ \hline A \supset C \end{array}$$

$$\begin{array}{l} \text{*b. } B \supset (A \ \& \ \neg B) \\ \hline \neg B \end{array}$$

$$\begin{array}{l} \text{c. } A = B \\ \quad \neg A \\ \hline \neg B \end{array}$$

$$\begin{array}{l} \text{*d. } A \supset (B \ \& \ C) \\ \quad \neg C \\ \hline \neg A \end{array}$$

$$\begin{array}{l} \text{e. } D \\ \hline A \supset [B \supset (C \supset D)] \end{array}$$

$$\begin{array}{l} \text{*f. } A = B \\ \quad B = C \\ \hline A = C \end{array}$$

$$\begin{array}{l} \text{g. } A \supset (B \supset C) \\ \quad D \supset B \\ \hline A \supset (D \supset C) \end{array}$$

$$\begin{array}{l} \text{*h. } \neg B \supset A \\ \quad C \vee \neg B \\ \hline \neg C \\ \hline A \end{array}$$

$$\begin{array}{l} \text{i. } \neg A \vee B \\ \quad B \supset C \\ \hline A \supset C \end{array}$$

$$\begin{array}{l} \text{*j. } (E \supset T) \ \& \ (T \supset O) \\ \quad O \supset E \\ \hline (E = O) \ \& \ (O = E) \end{array}$$

$$\begin{array}{l} \text{k. } A \supset (C \supset B) \\ \quad \neg C \supset \neg A \\ \hline A \\ \hline B \end{array}$$

$$\begin{array}{l} \text{*l. } \neg F \\ \quad \neg G \\ \hline \neg (F \vee G) \end{array}$$

$$\begin{array}{l} \text{m. } F = G \\ \quad F \vee G \\ \hline F \ \& \ G \end{array}$$

3. Prove that each of the following is a theorem in *SD*.

$$\text{a. } A \supset (A \vee B)$$

$$\text{*b. } A \supset (B \supset A)$$

$$\text{c. } A \supset [B \supset (A \ \& \ B)]$$

$$\text{*d. } (A \ \& \ B) \supset [(A \vee C) \ \& \ (B \vee C)]$$

$$\text{e. } (A = B) \supset (A \supset B)$$

$$\text{*f. } (A \ \& \ \neg A) \supset (B \ \& \ \neg B)$$

$$\text{g. } (A \supset B) \supset [(C \supset A) \supset (C \supset B)]$$

$$\text{*h. } (A \ \& \ B) \vee (\neg A \vee \neg B)$$

$$\text{i. } [(A \supset B) \ \& \ \neg B] \supset \neg A$$

$$\text{*j. } (A \ \& \ A) = A$$

$$\text{k. } A \supset [B \supset (A \supset B)]$$

$$\text{*l. } \neg A \supset [(B \ \& \ A) \supset C]$$

$$\text{m. } (A \supset B) \supset [\neg B \supset \neg (A \ \& \ D)]$$

$$\text{*n. } [\neg A \supset \neg (A \supset B)] \supset A$$

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gas mulia gas mulia adalah unsur unsur yang terdapat dalam golongan viiia yang memiliki kestabilan yang sangat tinggi dan sebagian ditemukan di alam dalam bentuk monoatomik unsur unsur yang terdapat dalam gas mulia yaitu helium he neon ne argon ar kripton kr xenon xe radon rn

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sifat sifat gas mulia wujud gas mulia unsur gas mulia terdapat sebagai gas tak berwarna yang monoatomik molekul yg terdiri dari 1 atom ini erat kaitannya dengan struktur elektron oktet dan duplet dari gas mulia

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dec 3 2015 gas mulia dan halogen merupakan unsur kimia yang sifatnya sangat bertolak belakang gas mulia dinamakan mulia karena seperti logam mulia sukar sekali bereaksi sementara halogen halos garam artinya adalah pembentuk garam halogen memiliki 6 bilangan oksidasi sehingga mudah sekali berreaksi dengan unsur lainnya gas mulia noble

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pembuatan gas mulia gas helium mempunyai titik didih yang sangat rendah yaitu 268 8 0c sehingga pemisahan gas helium dari gas alam dilakukan dengan cara pengembunan liquefaction sampai gas alam akan mencair sekitar 156 0c dan gas helium terpisah dari gas alam pembuatan ne ar kr dan xe dilakukan dengan proses destilasi bertingkat

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oct 7 2021 disebut gas mulia karena semua unsur pada golongan ini berwujud gas dan memiliki konfigurasi elektron yang sangat stabil sehingga akan sangat sulit untuk bereaksi dengan unsur lainnya unsur gas mulia antara lain adalah helium he neon ne argon ar kripton kr xenon xe dan radon rn

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nov 26 2015 1 wujud gas mulia unsur gas mulia terdapat sebagai gas tak berwarna yang monoatomik ini erat kaitannya dengan struktur elektronoktet dan duplet dari gas mulia sedangkan wujud gas pada suhu kamar disebabkan titik

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apr 16 2017 mempunyai titik didih dan titik leleh yang sangat rendah jari jari atom gas mulia sangat kecil mempunyai elektron valensi 8 khusus untuk helium elektron valensinya 2 sudah stabil sehingga sukar membentuk senyawa dengan unsur lain dalam tidak pernah di temukan senyawa gas mulia gas mulia terdapat sebagai molekul monoatomik

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may 18 2022 dikutip dari buku kimia unsur golongan gas mulia halogen alkali dan alkali tanah yang disusun oleh tantri ambarsari 2020 halogen adalah senyawa yang unsur unsurnya dapat bereaksi dengan logam untuk membentuk garam reaktivitas unsur halogen sangat besar sehingga unsur unsurnya sulit ditemukan di alam