

9 • Bonding & Molecular Structure

LEWIS STRUCTURE PRACTICE - 1

| | | |
|--|--|---|
| Ca^{2+} $[Ca]^{2+}$ | NO_3^- $\leftrightarrow \left[\begin{array}{c} \text{O}^- - \text{N} = \text{O} \\ \\ \text{O}^- \end{array} \right]^- \leftrightarrow$ | PCl_5 |
| C_2H_2 $\text{H} : \text{C} : : : \text{C} : \text{H}$ | S^{2-} $[\text{:}\ddot{\text{S}}\text{:}]^{2-}$ | NH_3 $\text{H} : \ddot{\text{N}} : \text{H}$ $ $ H |
| SO_4^{2-} $\left[\begin{array}{c} \text{O} \\ \text{:}\ddot{\text{O}}\text{:} \\ \text{:}\ddot{\text{S}}\text{:} \\ \text{:}\ddot{\text{O}}\text{:} \\ \text{O} \end{array} \right]^{2-}$ | SO_2 $\text{O}^- - \text{S} = \text{O} \leftrightarrow \text{O} = \text{S} - \text{O}^-$ | CaH_2 $\text{H} : \text{Ca} : \text{H}$ |



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LEWIS STRUCTURE PRACTICE - 2

| | | |
|--|---|--|
| CH_4 $\text{H} - \text{C} - \text{H}$ $ $ H | HCN $\text{H} : \text{C} : : : \text{N} :$ | K^+ $[K]^+$ |
| PO_4^{3-} $\left[\begin{array}{c} \text{O} \\ \text{:}\ddot{\text{O}}\text{:} \\ \text{:}\ddot{\text{P}}\text{:} \\ \text{:}\ddot{\text{O}}\text{:} \\ \text{O} \end{array} \right]^{3-}$ | O_3 $\text{O}^- - \text{O} = \text{O} \leftrightarrow \text{O} = \text{O} - \text{O}^-$ | SF_6 |
| Br^- $[\text{:}\ddot{\text{Br}}\text{:}]^-$ | AlH_3 $\text{H} : \text{Al} : \text{H}$ $ $ H | NH_4^+ $\left[\begin{array}{c} \text{H} \\ \text{H} : \text{N} : \text{H} \\ \\ \text{H} \end{array} \right]^+$ |

Lewis Structure Worksheet 1 Answer Key

James S. Lewis



Lewis Structure Worksheet 1 Answer Key:

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