

16. The no. of Neutrons in ^{14}N d) 9
 a) 6 b) 7 c) 8
17. The energy of one quantum of radiation is calculated by d) $E = h\nu$
 a) $E = h\nu$ b) $h = E\nu$ c) $h = \nu/E$

(08)

Q.2) Answer the followings.

- i) What are Physical classification of Matter?
- ii) Define the elements and explains?
- iii) Define Atomic Mass?
- iv) Write the chemical equation for combustion of propane?
- v) Write down the IUPAC names?
 $\text{CH}_3\text{--C}(\text{C}_2\text{H}_5)_2\text{--CH}_3$ $\text{C}(\text{CH}_3)_4\text{--CH}_3$
- vi) Define the law of conservation of mass?
- vii) If an element X has mass no. 11 and it has 6 neutrons then write its representation.
- viii) State Heisenberg Uncertainty Principle?

Section- B

(16)

Attempt any Eight

- Q.3) How will you convert Propane into @-nitro propane?
- Q.4) How will you convert ethane into ethyl chloride?
- Q.5) State the uses of alkane?
- Q.6) write the physical properties of alkanes?
- Q.7) What are alkane? how they are classified?
- Q.8) Find the no. of Moles (n) present in 72 grams of H_2O ?
- Q.9) Write down the Units of Measurements?
- Q.10) Differentiates between Isotopes and Isobars?
- Q.11) State and Explains Pauli's exclusion principle
- Q.12) State Hund's rule of maximum Multiplicity?
- Q.13) State the order of fillings Atomic Orbital's model

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