1st Semester Chemistry Final Exam Review

Identify the following reaction type AND correct products:

Sc + 3 AgOH -->

- a. combustion
- b. synthesis
- c. decomposition
- d. single replacement
- e. double replacement
- w. Ag + ScOH
- x. Ag + 2 Sc(OH)₂
- y. 3 Ag + Sc(OH)₃
- z. 3 Ag + 2 Sc(OH)_e
- Convert: 2.32 x 105 mg = 7 kg 2

III. b. & z. III. b. & v. IV. d. & z. W. Q. St. Z. VI. d. & v.

m. & w.

VIII e. & y. VIIII. c. & z.

DX. d. & x.

Calculate the energy for a quantum of light with a frequency of 8.44 x 1018 Hz. 3.

4. Determine the oxidation number for iodine in aluminum iodate.

- 5. REPORTING WITH THE CORRECT NUMBER OF SIGNIFICANT DIGITS, how many grams are in 4.011 moles of calcium chloride?
 - a) 445,14078 grams b) 445,140 grams
- c) 445.1 grams
- d) 445 grams

- e) 450 grams
- f) 400 grams
- 6 Calculate the frequency of light that has a wavelength of 210 nm.

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