

Honors Earth Science Study Guide

Answer the following questions on a separate sheet of paper.

1. Define these parts of the water cycle.
 - Accumulation – (reservoirs) places where water builds up
 - Condensation – gaseous water forms liquid water after cooling
 - Evaporation – liquid water forms gaseous water after heating
 - Precipitation – gaseous water condenses in the atmosphere, causing rain, hail, snow, sleet, etc.
 - Runoff – precipitation over land collects and flows to lower, larger bodies of water
 - Transpiration – plants lose water from their leaves through small pore-like openings
 - Groundwater – water stored in the ground in aquifers, reservoirs, caverns, etc.
2. What causes the temperature to decrease with altitude in the troposphere?
The sun heats the ground and the water on Earth's surface. As you move farther from these heat sources, the air becomes colder.
3. The amount of water in Earth's atmosphere: **changes/does not change. (Circle One)**
Does not
4. What causes the temperature to rise in the stratosphere?
The ozonosphere contains ozone, a molecule that absorbs harmful light from the sun. As it absorbs this light, the light energy is transformed into heat.
5. What is the composition of the atmosphere? (In terms of elements)
Mostly nitrogen (78%), with smaller parts oxygen (21%) and other gases (<1%)
6. In your own words, describe how air pressure decreases as you climb a mountain. Use the word density.
Air pressure decreases as you move up a mountain because the molecules have weight. Most of the molecules are drawn toward the lowest layer of Earth, so fewer and fewer escape to outer layers. Since the number of these molecules is so small, and the amount of space is the same, the density is low.
7. During thermal expansion, density decreases. Explain why using the following words: mass, volume, kinetic energy.
During thermal expansion, density decreases because the heat absorbed by molecules is kinetic energy (motion of molecules – we define temperature as the speed of molecules, hence "kinetic"). As the substance expands, mass doesn't change – only its volume increases. We are not adding mass to the sample. Since density is mass/volume, density goes down as volume goes up.
8. How is temperature different from thermal energy?
Temperature is a measure of the speed of molecules, and has no relation to the number of molecules in a sample. But, total thermal energy combines all kinetic energy of all molecules in a sample, meaning it is heavily reliant on the number of molecules.
9. Why does warm air rise?
Warm air rises because the gases experience thermal expansion. As they expand, their density decreases. Less dense materials float on top of other ones, so the warm air will rise through cooler, denser air.
10. Wind is the movement of air caused by a difference in _____.
Air pressure
11. Air pressure always (naturally) flows from areas of _____ pressure to _____ pressure.
High, low

Honors Earth Science Study Guide Answers

Oberlin College



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Concepts of Earth Science & Chemistry Parent Lesson Plan John Hudson Tiner, 2013-08-26 Concepts of Earth and Chemistry Course Description This is the suggested course sequence that allows one core area of science to be studied per semester You can change the sequence of the semesters per the needs or interests of your student materials for each semester are independent of one another to allow flexibility Semester 1 Earth Blending a creationism perspective of history with definitions of terms and identification of famous explorers scientists etc this book gives students an excellent initial knowledge of people and places encouraging them to continue their studies in depth Semester 2 Chemistry Chemistry is an amazing branch of science that affects us every day yet few people realize it or even give it much thought Without chemistry there would be nothing made of plastic there would be no rubber tires no tin cans no televisions no microwave ovens or something as simple as wax paper This book presents an exciting and intriguing tour through the realm of chemistry as each chapter unfolds with facts and stories about the discoveries of discoverers Find out why pure gold is not used for jewelry or coins Join Humphry Davy as he made many chemical discoveries and learn how they shortened his life See how people in the 1870s could jump over the top of the Washington Monument Exploring the World of Chemistry brings science to life and is a wonderful learning tool with many illustrations and biographical information

Resources in Education, 1997

Discourse Analytic Perspectives on STEM Education Juliet Langman, Holly Hansen-Thomas, 2017-05-23 This volume explores the nature of discourse in secondary and upper elementary mathematics and science classrooms Chapters examine conditions that support or hinder teachers and students in particular language learners in employing language as a tool for learning The volume provides rich oral and written language examples from a range of classroom contexts to illustrate how linguistic practices affect students appropriation and display of disciplinary specific knowledge Chapters further explore linguistic practices through with the support of discourse analytic models that foreground the authentic classroom data with the aim of understanding the dynamics of the classroom The authors investigate the intersection between discourse and learning from a range of

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