#### Name:

#### Honors Chem Test: Ch. 7

		Choice—1 point each e choice that best completes the statemen	et or answer	s the question	
	1.	How many atoms of fluorine are present in a molecule of carbon tetrafluonde, CF.7			
		a 1	e.	4	
		b. 2	d.	5	
	2.	Name the compound Al <sub>2</sub> S <sub>3</sub>			
		a. aluminum sulfate	· C	altuminum(II) sulfate	
		b. aluminum sulfur	di.	aluminum sulfide	
	3.	Name the compound CF.			
		a. calcium fluoride	c.	carbon tetrafluonide	
		b. carbon fluoride	d.	mono-calcium quadraffuoride	
	4.	Name the compound $N_2O_3$			
		a. dimitrogen omde	C	natri e con de	
		<ul> <li>nitrogen triconde</li> </ul>	d.	dimitrogen triconde	
	51	What is the formula for diasrsenic trisulfide?			
		a. Ar <sub>2</sub> S <sub>2</sub>	C.	As <sub>2</sub> S <sub>4</sub>	
		b. ArSı	d.	As <sub>3</sub> S <sub>3</sub>	
	6.	What is the formula for sulfur dichloride?			
		a. SCI	(C)	S <sub>2</sub> C1	
		b SCI;	d.	S <sub>2</sub> Cl <sub>2</sub>	
	7.	What is the formula for hydrochloric a	rcid?		
		a. HF	C.	HCIO	
		ь НСІ	d.	H <sub>2</sub> CO <sub>2</sub>	
	8. The molar mass of an element is the mass of one				
		a atom of the element.	c.	gram of the element.	
		<ul> <li>b. crystal of the element.</li> </ul>	d.	mole of the element.	
	9.	The molar mass of Mgl <sub>2</sub> is			
		a. the sum of the masses of 1 mol of Mg and 2 mol of 1.			
		<ul> <li>the sum of the masses of 1 mol of Mg and 1 mol of I.</li> </ul>			
		c the sum of the masses of 1 atom of Mg and 2 atoms of I.			
		d. the sum of the masses of 1 atom of Mg and 1 atom of I.			

# **Honors Chemistry Practice Test Multiple Choice Thermo**

**Anthony S. Fauci** 

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### **Table of Contents Honors Chemistry Practice Test Multiple Choice Thermo**

- 1. Understanding the eBook Honors Chemistry Practice Test Multiple Choice Thermo
  - The Rise of Digital Reading Honors Chemistry Practice Test Multiple Choice Thermo
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Honors Chemistry Practice Test Multiple Choice Thermo
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Honors Chemistry Practice Test Multiple Choice Thermo
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Honors Chemistry Practice Test Multiple Choice Thermo
  - Personalized Recommendations
  - Honors Chemistry Practice Test Multiple Choice Thermo User Reviews and Ratings
  - Honors Chemistry Practice Test Multiple Choice Thermo and Bestseller Lists
- 5. Accessing Honors Chemistry Practice Test Multiple Choice Thermo Free and Paid eBooks
  - Honors Chemistry Practice Test Multiple Choice Thermo Public Domain eBooks
  - Honors Chemistry Practice Test Multiple Choice Thermo eBook Subscription Services
  - Honors Chemistry Practice Test Multiple Choice Thermo Budget-Friendly Options

- 6. Navigating Honors Chemistry Practice Test Multiple Choice Thermo eBook Formats
  - o ePub, PDF, MOBI, and More
  - Honors Chemistry Practice Test Multiple Choice Thermo Compatibility with Devices
  - Honors Chemistry Practice Test Multiple Choice Thermo Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - o Adjustable Fonts and Text Sizes of Honors Chemistry Practice Test Multiple Choice Thermo
  - Highlighting and Note-Taking Honors Chemistry Practice Test Multiple Choice Thermo
  - Interactive Elements Honors Chemistry Practice Test Multiple Choice Thermo
- 8. Staying Engaged with Honors Chemistry Practice Test Multiple Choice Thermo
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Honors Chemistry Practice Test Multiple Choice Thermo
- 9. Balancing eBooks and Physical Books Honors Chemistry Practice Test Multiple Choice Thermo
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Honors Chemistry Practice Test Multiple Choice Thermo
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Honors Chemistry Practice Test Multiple Choice Thermo
  - Setting Reading Goals Honors Chemistry Practice Test Multiple Choice Thermo
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Honors Chemistry Practice Test Multiple Choice Thermo
  - Fact-Checking eBook Content of Honors Chemistry Practice Test Multiple Choice Thermo
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements

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