



**DS ELECTRIC
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MANUAL NUMBER
1014187
ISSUE DATE SEPTEMBER 1991
0991D0881A

Golf Club Car Owners Manual

S Marginson



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web the half life of a specific radioactive isotope is constant it is unaffected by conditions and is independent of the initial amount of that isotope consider the following example suppose we have 100 0 g of tritium a radioactive isotope of hydrogen

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disintegrates over a period of 90 years how many g of cesium 137 would remain a we r 2 actinium 226 has a half life of 29 hours

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web we utilize the equation that relate amount remaining initial mass and number of half lives $n = \frac{t}{T} \times n_0$ 124 x 503

125 g 5 what is the half life of an isotope that is 80 remained after 16 days solution remaining 80 100 therefore $n = \frac{t}{T} \times n_0$ 100

now using the half life equation in example 1 we have

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web section learning objectives by the end of this section you will be able to do the following explain radioactive half life and its role in radiometric dating calculate radioactive half life and solve problems associated with radiometric dating section key terms half life and the rate of radioactive decay unstable nuclei decay

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