

Name: \_\_\_\_\_ Date: \_\_\_\_\_ Lab Section \_\_\_\_\_

## Experiment 2: Molecular Models

### REPORT SHEET

### Pre-lab Study Questions:

1. Objectives of the experiment (in your own words):
2. What kinds of electrons form the covalent bonds in molecules?
3. Is a polar molecule symmetrical or asymmetrical?

FORMULA OF MOLECULE	LEWIS STRUCTURE	3-D SHAPE w/lone pairs	MOLECULAR GEOMETRY	MOLECULAR POLARITY
example: $\text{CH}_4$			tetrahedral	nonpolar
$\text{H}_2\text{O}$			tetrahedral	polar
$\text{H}_2\text{O}^+$			tetrahedral	nonpolar
$\text{C}_2\text{H}_6$				
$\text{C}_2\text{H}_4$			trigonal planar	Non polar
$\text{C}_2\text{H}_2$				
$\text{CO}_2$			Linear	Non polar
$\text{H}_2\text{O}_2$				
$\text{CH}_3\text{Cl}$				

# Geometrical Structure Of Molecules Lab Answers

**PT Brinkman**



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