

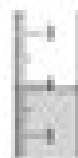
7. A student investigated the reaction between dilute hydrochloric acid and two different aqueous solutions of sodium hydroxide (solution Q) and solution P.

Two experiments were carried out.

**Experiment 1**

1. A burette was filled with dilute hydrochloric acid. The initial burette reading was recorded.
2. Using a measuring cylinder, 25.0 cm<sup>3</sup> of solution Q were poured into a conical flask.
3. Thymolphthalein indicator was added to the contents of flask.
4. The dilute hydrochloric acid was added from the burette, while swirling the flask, until the solution just changed colour. The final burette reading was recorded.

- (a) Use the burette diagrams to record the readings in the table and complete the table.



initial reading



final reading

final burette reading / cm <sup>3</sup>	
initial burette reading / cm <sup>3</sup>	
difference / cm <sup>3</sup>	

(1)

**Experiment 2**

1. The conical flask was emptied and rinsed with distilled water.
2. Experiment 1 was repeated using solution P instead of solution Q.

- (a) Use the burette diagrams to record the readings in the table and complete the table.



initial reading



final reading

final burette reading / cm <sup>3</sup>	
initial burette reading / cm <sup>3</sup>	
difference / cm <sup>3</sup>	

(1)

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**Huangqi Zhang**

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