

PHYSICAL SCIENCES (CHEMISTRY) GRADE 11 TERM 3 FORMAL EXPERIMENT - PART 2

TIME: 1 HOUR

(1)

(5)

(6)

VERIFICATION OF BOYLE'S LAW

INDIVIDUAL WORK (should be done under strict supervision).

- Write a scientific report which must include the following: 100
- 1.1 1.2 Apparatus and method
- 2. Observations and Recording of results (Copy the table below onto your book or sheet of paper and transfer the results you recorded when conducting the experiment). Calculate and pV for each set of results and record your answers in the table.

Volume (V) cm³	Pressure (p) kPa	pV (*)
	-	

- Analysis of the results:
- 3.1 Identify the following variables for the experiment:
 - Independent
 - Dependent
- Controlled (3)3.2 Use the graph paper provided by your teacher to draw graphs of V versus p and
- V versus . (18).
- 4 Conclusion
 - 4.1 Examine the results in your table as well as the two graphs you have drawn, what happens to the volume of the gas in the tube as the pressure is decreased. (1)
 - 4.2 Write down a conclusion for this experiment. (3)

55 Questions and discussion:

- 5.1 Explain why you have to wait for a few minutes before you can take readings.
- (2)5.2 Explain the relationship between pressure and volume in terms of the kinetic theory of gases. (4)

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Table of Contents Mpumalanga Grade 11 2013 Physics Papers

- 1. Understanding the eBook Mpumalanga Grade 11 2013 Physics Papers
 - The Rise of Digital Reading Mpumalanga Grade 11 2013 Physics Papers
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Mpumalanga Grade 11 2013 Physics Papers
 - Exploring Different Genres
 - o Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Mpumalanga Grade 11 2013 Physics Papers
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Mpumalanga Grade 11 2013 Physics Papers
 - Personalized Recommendations
 - $\circ\,$ Mpumalanga Grade 11 2013 Physics Papers User Reviews and Ratings
 - Mpumalanga Grade 11 2013 Physics Papers and Bestseller Lists
- 5. Accessing Mpumalanga Grade 11 2013 Physics Papers Free and Paid eBooks
 - Mpumalanga Grade 11 2013 Physics Papers Public Domain eBooks
 - Mpumalanga Grade 11 2013 Physics Papers eBook Subscription Services
 - Mpumalanga Grade 11 2013 Physics Papers Budget-Friendly Options

- 6. Navigating Mpumalanga Grade 11 2013 Physics Papers eBook Formats
 - o ePub, PDF, MOBI, and More
 - Mpumalanga Grade 11 2013 Physics Papers Compatibility with Devices
 - Mpumalanga Grade 11 2013 Physics Papers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Mpumalanga Grade 11 2013 Physics Papers
 - Highlighting and Note-Taking Mpumalanga Grade 11 2013 Physics Papers
 - Interactive Elements Mpumalanga Grade 11 2013 Physics Papers
- 8. Staying Engaged with Mpumalanga Grade 11 2013 Physics Papers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Mpumalanga Grade 11 2013 Physics Papers
- 9. Balancing eBooks and Physical Books Mpumalanga Grade 11 2013 Physics Papers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Mpumalanga Grade 11 2013 Physics Papers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Mpumalanga Grade 11 2013 Physics Papers
 - Setting Reading Goals Mpumalanga Grade 11 2013 Physics Papers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Mpumalanga Grade 11 2013 Physics Papers
 - Fact-Checking eBook Content of Mpumalanga Grade 11 2013 Physics Papers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements

• Interactive and Gamified eBooks

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