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by radioactive dating unstable nuclei decay however some nuclides decay faster than others for example radium and polonium discovered by the curies decay faster than uranium

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web we utilize the equation that relate amount remaining initial mass and number of half lives $n = \frac{t}{t_{1/2}} \times \ln 2$
 $\ln 2 = 0.693$
 $n = \frac{16 \text{ days}}{125 \text{ g}} \times 0.693$
 $n = 0.086$
 $n = 0.086 \times 125 \text{ g} = 10.75 \text{ g}$
 $125 \text{ g} - 10.75 \text{ g} = 114.25 \text{ g}$
 125 g 5 what is the half life of an isotope that is 80 remained after 16 days solution remaining 80 100 therefore $n = \frac{t}{t_{1/2}} \times \ln 2$
 $n = \frac{16 \text{ days}}{80} \times 0.693$
 $n = 0.1386$
 $n = 0.1386 \times 100 = 13.86 \text{ g}$
 now using the half life equation in example 1 we have

web what is the half life of this radioisotope after 2 days 5 grams of the radioisotope 1 day b 2 days c 5 days d 4 days 3 the ratio of uranium 238 to lead 206 in a mineral is used to determine age b density c solubility d composition 4 a radioactive isotope has a half life of 10 years

web chapter 15 nuclear chemistry half life learning objectives define half life determine the amount of radioactive substance remaining after a given number of half lives whether or not a given isotope is radioactive is a characteristic of that particular isotope

web the half life of a radioactive isotope is the amount of time it takes for one half of the radioactive isotope to decay the half life of a specific radioactive isotope is constant it is unaffected by conditions and is independent of the initial amount of that isotope

web section learning objectives by the end of this section you will be able to do the following explain radioactive half life and its role in radiometric dating calculate radioactive half life and solve problems associated with radiometric dating section key terms half life and the rate of radioactive decay unstable nuclei decay

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web jul 24 2023 directions common core state standards printable worksheets and classroom aids objectives students will learn about radioactive decay and decay chains demonstrate the concept of half life calculate and chart the half life of a given sample discuss the significance of knowing the half life of radioactive elements

web solution if we compare the time that has passed to the isotope's half life we note that 44 0 s is exactly 4 half lives so

using the previous equation n 4 substituting and solving results in the following less than one third of a gram of ^{20}F remains
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