

59. A) Na^+, Cl^-
B) K^+, Cl^-
C) $\text{Na}^+, \text{Mg}^{2+}$
D) Na^+, F^-
Which one of the following hydrogen bonds is stronger than others?
A) $\text{N}^{\delta-} - \text{H}^{\delta+} \cdots \text{N}^{\delta-} - \text{H}^{\delta+}$
B) $\text{F}^{\delta-} - \text{H}^{\delta+} \cdots \text{F}^{\delta-} - \text{H}^{\delta+}$
C) $\text{O}^{\delta-} - \text{H}^{\delta+} \cdots \text{O}^{\delta-} - \text{H}^{\delta+}$
D) $\text{N}^{\delta-} - \text{H}^{\delta+} \cdots \text{O}^{\delta-} - \text{H}^{\delta+}$
60. The half life of N_2O_5 at 40°C is 24 minutes. How long will it take for sample of N_2O_5 to decay to 25% of its original concentration?
A) 24 minutes
B) 72 minutes
C) 120 minutes
D) 48 minutes
61. When the change in concentration is $6 \times 10^{-4} \text{ mol dm}^{-3}$ and time for that change is 10 seconds, the rate of reaction will be
A) $6 \times 10^{-3} \text{ mol dm}^{-3} \text{ sec}^{-1}$
B) $6 \times 10^{-4} \text{ mol dm}^{-3} \text{ sec}^{-1}$
C) $6 \times 10^{-2} \text{ mol dm}^{-3} \text{ sec}^{-1}$
D) $6 \times 10^{-5} \text{ mol dm}^{-3} \text{ sec}^{-1}$
62. Which one of the following will have the smallest radius?
A) Al^{3+}
B) Si^{4+}
C) Mg^{2+}
D) Na^{+1}
63. Keeping in view the size of atoms, which order is correct?
A) $\text{N} > \text{C}$
B) $\text{P} > \text{Si}$
C) $\text{Ar} > \text{Cl}$
D) $\text{Li} > \text{Be}$
64. On the basis of oxidizing power of halogens, which reaction is possible?
A) $\text{I}_2 + 2\text{Cl}^- \longrightarrow \text{Cl}_2 + 2\text{I}^-$
B) $\text{Br}_2 + 2\text{I}^- \longrightarrow \text{I}_2 + 2\text{Br}^-$
C) $\text{Cl}_2 + 2\text{F}^- \longrightarrow \text{F}_2 + 2\text{Cl}^-$
D) $\text{I}_2 + 2\text{Br}^- \longrightarrow \text{Br}_2 + 2\text{I}^-$
65. Which one of the following gases is used as mixture for breathing by sea divers?
A) Oxygen and Nitrogen
B) Nitrogen and Helium
C) Helium and Oxygen
D) Helium and Hydrogen
66. $[\text{Ti}(\text{H}_2\text{O})_6]^{3+}$ transmits
A) Yellow and red light
B) Yellow and blue light
C) Red and white light
D) Red and blue light
67. Electronic configuration of Gold $[\text{Au}_{79}]$ is
A) $[\text{Xe}]4f^{14}, 5d^{10}, 6s^1$
B) $[\text{Xe}]4f^{10}, 5d^{10}, 6s^2$
C) $[\text{Xe}]4f^{14}, 5d^9, 6s^2$
D) $[\text{Xe}]4f^{14}, 5d^{10}, 6s^2$
68. About 80% of ammonia is used for the production of
A) Explosives
B) Fertilizers
C) Nylon
D) polymers
69. Urea is the most widely used nitrogen fertilizer in Pakistan. Its composition is
A) NH_2CO
B) $\text{N}_2\text{H}_5\text{CO}_2$
C) $\text{N}_2\text{H}_4\text{CO}_2$
D) $\text{N}_2\text{H}_4\text{CO}$
70. During the manufacture of nitric acid, nitric oxide is oxidized to nitrogen dioxide. This reaction is given as
$$2\text{NO}_{(g)} + \text{O}_{2(g)} \rightleftharpoons 2\text{NO}_{2(g)} \quad \Delta H = -114 \text{ kJ/mol}$$

According to Le Chatelier's Principle
A) Reaction must not be temperature dependent
B) Reaction must be carried out at room temperature
C) Reaction must be carried out at low temperature

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