

1. (a) (i) electrons ...16..... protons .....16 ... neutrons ...16....  
**All correct (1)** 1
- (ii) they are isotopes (1)  
 The word 'isotope' or some derivative is essential 1
- (iii)  $S^{2-}$  / sulphide(2-) **ion** / sulphur -2 ion / sulphur with two extra electrons (1)  
 but **not** just S or sulphur 1
- (b) (i)  $X(g) + e^- \rightarrow X^-(g)$  ( $S(g) + e^- \rightarrow S^-(g)$ ) (1)  
 $X^-(g) + e^- \rightarrow X^{2-}(g)$  ( $S^-(g) + e^- \rightarrow S^{2-}(g)$ ) (1)  
 can have **Z** in place of **X<sup>2-</sup>** but not **Z<sup>2-</sup>**  
 Penalize omission of state symbol (g) only once. 2
- (ii) adding a **negative onto a negative species** requires energy to overcome repulsion (1)  
**or**  
 second electron repelled by negative charge  
**Not** second electron repelled by first 1
- (c) (Bdecay from) **Y produces only one isotope( <sup>39</sup>Cl)** (1)  
 natural sample contains two isotopes (35 and 37) (1) 2

[8]

2. (a) (i) magnesium / lead / iron (1) 1
- (ii) iron (1) 1
- (iii) carbon / phosphorus (1) 1
- (iv) magnesium / lead (1) 1
- (v) carbon / lead (1) 1
- (b) Helium (1)  
 because it is the **smallest atom** (1)  
 with an **unshielded nucleus** (1)  
 Hydrogen can score max 1 for 'unshielded nucleus'  
 If candidate chooses Carbon, i.e. uses list from part (a),  
 can score 1 mark for reference to 'smallest atom' 3
- (c) (i) ability of a positive ion / cation to attract electrons / distort / pull the electron cloud (1)  
 around an anion (1) 2
- (ii) size (1) **Not** atomic radius  
 charge (1)  
 charge density **worth** (2)  
 volume of ion can score in place of size 2
- (iii) magnesium chloride  
 magnesium ion smaller than calcium ion (1)  
 polarizes / distorts / attracts electrons around chloride more (1) 2

[14]

# June 2013 Chemistry Unit 5 Mark Scheme

**Christian Drosten**



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