

Name Key Class \_\_\_\_\_ Date \_\_\_\_\_

## Acid-Base Theories

1. Summarize the three main acid-base theories in the table below.

|                | ACID                                                                | BASE                                       |
|----------------|---------------------------------------------------------------------|--------------------------------------------|
| Arrhenius      | $\text{H}^+$ or $\text{H}_3\text{O}^+$ producer in aqueous solution | $\text{OH}^-$ producer in aqueous solution |
| Bronsted-Lowry | proton ( $\text{H}^+$ ) donor                                       | proton ( $\text{H}^+$ ) acceptor           |
| Lewis          | electron-pair acceptor                                              | electron-pair donor                        |

Use the following chemical equations to answer question 2.



2. According to Arrhenius, HCl would be classified as a(n)
- acid
- and
- $\text{NH}_3$
- would be classified as a(n)
- base
- . Explain why.

HCl produces hydronium ions ( $\text{H}_3\text{O}^+$ ) in water.  $\text{NH}_3$  produces hydroxide ions ( $\text{OH}^-$ ) in water.For problems 3 – 5, label the acid (A), base (B), conjugate acid (CA), and conjugate base (CB) in each of the following reactions. Show the transfer of the proton ( $\text{H}^+$ ) by drawing an arrow.

For problems 6 – 9, give the conjugate base for each Bronsted-Lowry acid.



For problems 10 – 13, give the conjugate acid for each Bronsted-Lowry base.



# Holt Chemistry Acid And Bases Review Answers

**William L. Ramsey**



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**Instructor's Manual for General Chemistry, Experiment and Theory** Richard H. Eastman, 1970      *Modern Chemistry* Nicholas D. Tzimopoulos, 1993      *The Science Teacher*, 1968      SCC Library has 1964 cur      **Monograph Series**, 1928

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In an electronic era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Holt Chemistry Acid And Bases Review Answers**," compiled by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we shall delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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