

**NEB CLASS 11**  
**MODEL QUESTION**  
**(EFFECTIVE FROM 2077)**

Grade XI

Subject Code : Che. 301

Time :3 hours

Attempt all questions

Subject: Chemistry (Theory)

Full marks: 75

**Group 'A'**

Circle the best alternative to the following questions.

(11×1 = 11)

1. How many atoms are there in two molecules of water?
  - a. 3
  - b. 4
  - c. 5
  - d. 6
2. What is the number of moles of ammonia gas formed when 0.5 mole of nitrogen gas is reacted with excess of hydrogen gas?
  - a. 0.5
  - b. 1
  - c. 2
  - d. 3
3. Which of the following bonding is responsible for the solubility of ammonia gas in water?
  - a. Hydrogen bonding
  - b. Ionic bonding
  - c. Covalent bonding
  - d. Van der Waals' force
4. What happens when Sulphur dioxide ( $\text{SO}_2$ ) gas is passed through an acidified solution of hydrogen sulfide ( $\text{H}_2\text{S}$ ) gas?
  - a.  $\text{SO}_2$  is oxidized to Sulphur
  - b.  $\text{H}_2\text{S}$  is reduced to Sulphur
  - c.  $\text{SO}_2$  is oxidized to  $\text{H}_2\text{SO}_4$
  - d.  $\text{SO}_2$  is reduced to Sulphur

# Grade11 Chemistry Question Paper

**Annelies Wilder-Smith**



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## **Table of Contents Grade11 Chemistry Question Paper**

1. Understanding the eBook Grade11 Chemistry Question Paper
  - The Rise of Digital Reading Grade11 Chemistry Question Paper
  - Advantages of eBooks Over Traditional Books
2. Identifying Grade11 Chemistry Question Paper
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Grade11 Chemistry Question Paper
  - User-Friendly Interface
4. Exploring eBook Recommendations from Grade11 Chemistry Question Paper
  - Personalized Recommendations
  - Grade11 Chemistry Question Paper User Reviews and Ratings
  - Grade11 Chemistry Question Paper and Bestseller Lists
5. Accessing Grade11 Chemistry Question Paper Free and Paid eBooks
  - Grade11 Chemistry Question Paper Public Domain eBooks
  - Grade11 Chemistry Question Paper eBook Subscription Services
  - Grade11 Chemistry Question Paper Budget-Friendly Options
6. Navigating Grade11 Chemistry Question Paper eBook Formats

- ePub, PDF, MOBI, and More
- Grade11 Chemistry Question Paper Compatibility with Devices
- Grade11 Chemistry Question Paper Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Grade11 Chemistry Question Paper
  - Highlighting and Note-Taking Grade11 Chemistry Question Paper
  - Interactive Elements Grade11 Chemistry Question Paper
- 8. Staying Engaged with Grade11 Chemistry Question Paper
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Grade11 Chemistry Question Paper
- 9. Balancing eBooks and Physical Books Grade11 Chemistry Question Paper
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Grade11 Chemistry Question Paper
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Grade11 Chemistry Question Paper
  - Setting Reading Goals Grade11 Chemistry Question Paper
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Grade11 Chemistry Question Paper
  - Fact-Checking eBook Content of Grade11 Chemistry Question Paper
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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