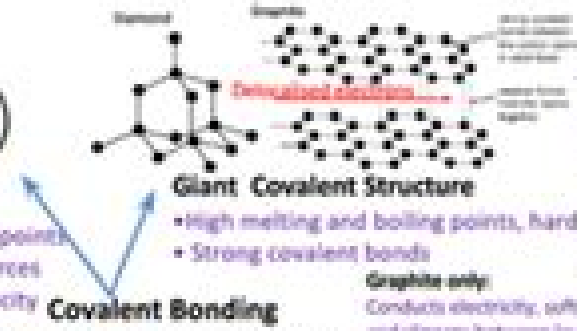
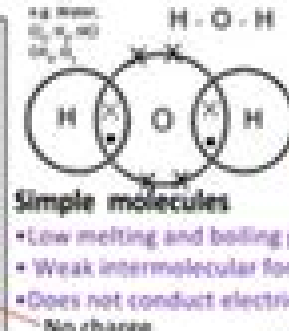




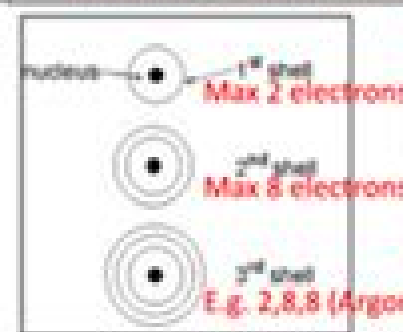
Periodic Table of the Elements

An atom:
Number of Protons = Number of Electrons
Mass number = Protons + Neutrons

C2 Chemistry

Periodic table arranges elements by **ATOMIC NUMBER** (proton number/the small one)

The number of outer shell electrons match the group the element is found in.
E.g. Lithium 2,1 is a group 1 element.

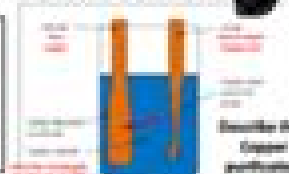


Protons and neutrons both have a mass of 1

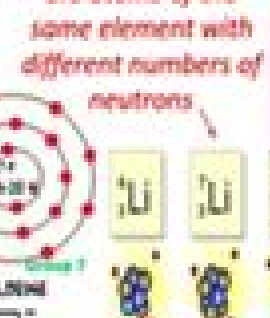
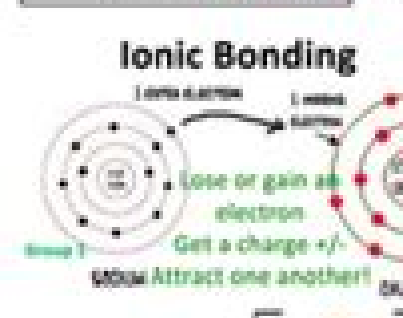
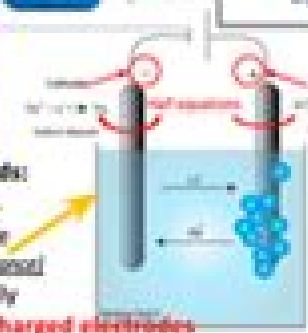
ISOTOPES
are atoms of the same element with different numbers of neutrons

OIL RIG

Oxidation is loss (of electrons)
Reduction is gain (of electrons)



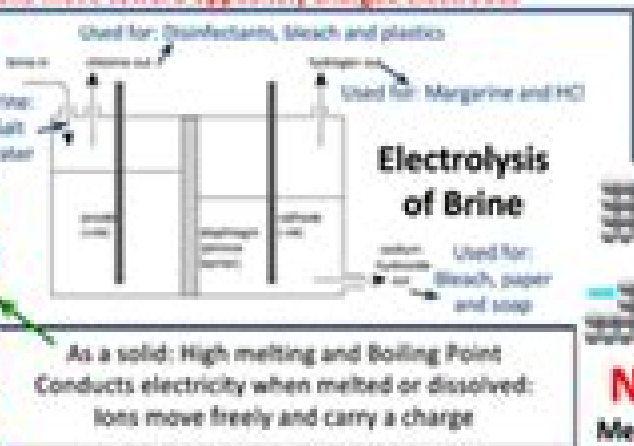
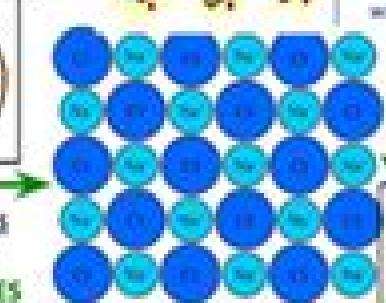
Electrolysis:
Splitting up a substance using electricity



When there are 2 possible products
The less reactive forms at the electrode

Electrolysis needs:

- 2 electrodes
- An electrolyte (ionic or molten substance)
- A power supply



Metal ions only arranged in layers
ELECTROSTATIC ATTRACTION
between
+IONS and ELECTRONS

Metal ions slide over one another making them **MALLEABLE**

Metals conduct **HEAT** well

NANO-TECHNOLOGY
Means **REALLY REALLY REALLY** small!!!!

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J. Bennetts, M. Hannon, J. Mundie



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Table of Contents Gcse Edexcel Chemistry C2 Past Papers

1. Understanding the eBook Gcse Edexcel Chemistry C2 Past Papers
 - The Rise of Digital Reading Gcse Edexcel Chemistry C2 Past Papers
 - Advantages of eBooks Over Traditional Books
2. Identifying Gcse Edexcel Chemistry C2 Past Papers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Gcse Edexcel Chemistry C2 Past Papers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Gcse Edexcel Chemistry C2 Past Papers
 - Personalized Recommendations
 - Gcse Edexcel Chemistry C2 Past Papers User Reviews and Ratings
 - Gcse Edexcel Chemistry C2 Past Papers and Bestseller Lists
5. Accessing Gcse Edexcel Chemistry C2 Past Papers Free and Paid eBooks
 - Gcse Edexcel Chemistry C2 Past Papers Public Domain eBooks
 - Gcse Edexcel Chemistry C2 Past Papers eBook Subscription Services
 - Gcse Edexcel Chemistry C2 Past Papers Budget-Friendly Options
6. Navigating Gcse Edexcel Chemistry C2 Past Papers eBook Formats
 - ePub, PDF, MOBI, and More
 - Gcse Edexcel Chemistry C2 Past Papers Compatibility with Devices
 - Gcse Edexcel Chemistry C2 Past Papers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Gcse Edexcel Chemistry C2 Past Papers
 - Highlighting and Note-Taking Gcse Edexcel Chemistry C2 Past Papers
 - Interactive Elements Gcse Edexcel Chemistry C2 Past Papers
8. Staying Engaged with Gcse Edexcel Chemistry C2 Past Papers

- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Gcse Edexcel Chemistry C2 Past Papers
9. Balancing eBooks and Physical Books Gcse Edexcel Chemistry C2 Past Papers
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Gcse Edexcel Chemistry C2 Past Papers
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Gcse Edexcel Chemistry C2 Past Papers
- Setting Reading Goals Gcse Edexcel Chemistry C2 Past Papers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Gcse Edexcel Chemistry C2 Past Papers
- Fact-Checking eBook Content of Gcse Edexcel Chemistry C2 Past Papers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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