

Explain how you identified the limiting and excess reactant based only on your qualitative observations (not quantitative measurements or calculations). Type your answer into the box. Use complete sentences.

According to observation the initial color of bromocresol purple is yellow in color after adding the NaOH, it turns violet. After Changing the color yellow from purple indicates acid is completely consumed so excess of base addition changes the color into violet.

**Table C. Calculations to Determine Theoretical Yields**

<i>Description</i>	<i>Calculated Value</i>
Volume of NaOH solution	22.20 mL
Initial moles of NaOH	0.002451 moles
Initial moles of $\text{H}_3\text{PO}_4$	0.002602 moles
Mass of $\text{H}_2\text{O}$ produced if NaOH runs out first	0.04414 g
Mass of $\text{H}_2\text{O}$ produced if $\text{H}_3\text{PO}_4$ runs out first	1.265 g
Theoretical yield of $\text{H}_2\text{O}$ (in grams)	1.265 g
Theoretical yield of $\text{Na}_3\text{PO}_4$ (in grams)	0.4267 g

**Table D. Identification of Limiting and Excess Reactants from Calculations**

<i>Description</i>	<i>Reactant Name or Formula</i>
Limiting Reactant	Phosphoric Acid
Excess Reactant	Sodium Hydroxide

Explain how you identified the limiting and excess reactant based on your calculations. Be specific. Type your answer into the box. Use complete sentences.

According to our calculation the reacting equivalent of  $\text{H}_3\text{PO}_4$  is less than NaOH. Therefore  $\text{H}_3\text{PO}_4$  is limiting reactant so product formation depends on  $\text{H}_3\text{PO}_4$  because phosphoric acid is completely consumed.

# Formal Lab Report On Limiting Reactant

**Marco Cascella**



## Formal Lab Report On Limiting Reactant:

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*Working with Chemistry* Donald J. Wink, Sharon Fetzer-Gislason, Julie Ellefson Kuehn, 2004-02-20 With this modular laboratory program students build skills using important chemical concepts and techniques to the point where they are able to design a solution to a scenario drawn from a professional environment The scenarios are drawn from the lives of people who work with chemistry every day ranging from field ecologists to chemical engineers and include many health professionals as well *Experimental Organic Chemistry* Joaquín Isac-García, José A. Dobado, Francisco G.

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Richard B. Friedman,Donald S. Young,1989 An aid to determine the possible cause of laboratory test abnormalities encountered in clinical practice Sections include laboratory test index disease keyword index laboratory test listings disease listings by ICD 9CM classification and references

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Suthan S. Suthersan,John Horst,Matthew Schnobrich,Nicklaus Welty,Jeff McDonough,2016-11-25 Remediation engineering has evolved and advanced from the stage of being a sub discipline of environmental engineering into its own engineering discipline supporting the growth of a global industry This fully updated second edition will capture the fundamental advancements that have taken place during the last two decades within the sub disciplines that form the foundation of the remediation engineering platform The book will cover the entire spectrum of current technologies that are being employed in this industry and will also touch on future trends and how practitioners should anticipate and adapt to those needs

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