



GENERAL MATHEMATICS

1. The length of a rectangular field is seven times its breadth. If the area of the field be 252 m^2 . Then the length is
(A) 40 m
(B) 42 m
(C) 44 m
(D) 50 m
2. Split 23 into two parts such that the difference of the squares of each part is also 23. Then the two parts are
(A) 10, 13
(B) 7, 16
(C) 11, 12
(D) 8, 15

सामान्य मणित

1. एक आयताकार मैदान की लम्बाई उसकी चौड़ाई से 7 गुना ज्यादा है। यदि मैदान का क्षेत्रफल 252 वर्ग मीटर हो, तो लम्बाई होगी
- (A) 40 मीटर
(B) 42 मीटर
(C) 44 मीटर
(D) 50 मीटर
2. 23 को दो भागों में इस प्रकार विभाजित करो कि दो भागों के वर्गों का अंतर पुनः 23 हो, तो वह दो भाग होंगे
- (A) 10, 13
(B) 7, 16
(C) 11, 12
(D) 8, 15
- Handwritten calculations and markings:
- For Q1: $144 \div 13$
 - For Q2: 121 and 100 (with 100 crossed out and 121 written above it)

$$A \oplus B = 203$$

$$\begin{array}{r} 121 \\ 149 \\ \hline 13 \end{array}$$

$$\begin{array}{r} 2 \\ \hline A + B = 23 \\ A + B = 23 \\ \hline 2A = 46 \end{array}$$

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Lei Huang



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web he concluded that at a constant pressure the volume of the gas would expand at a particular rate for each degree of temperature rise that being 1 267 per degree in 1874 victor regnault obtained better experimental results showing this number to be 1 273 and concluded that the pressure would approach zero at 1 273 degrees c

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web the volume of 1.00 mol of any gas at stp standard temperature 273.15 K and pressure 1 atm is measured to be 22.414 L we can substitute 101.325 kPa for pressure 22.414 L for volume and 273.15 K for temperature into the ideal gas equation and solve for r

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web chemists sometimes make comparisons against a standard temperature and pressure stp for reporting properties of gases 273.15 K and 1 atm 101.325 kPa 1 atm stp one mole of an ideal gas has a volume of about 22.4 L this is referred to as the standard molar volume figure 9.18

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web the pressure volume temperature pvt fluids lab in KU's chemical petroleum engineering department is used to study the properties of fluids under a wide range of pressure and temperature

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web volume pressure kPa linear pressure kPa volume mL pressure kPa in part b we observed the relationship between temperature and pressure the pressure of the 125 mL flask was at 101.36 kPa when the temperature was at 23.2 °C

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82 kpa 16 0 c 289 k 0 33 97 90 kpa 19 0 c 292 k 0 34 98 97 kpa 99 22 kpa 101 4 kpa 19 1 c

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