

Lab 10: Chapter 1 - Laboratory Techniques: Measurement Lab Report

Instructions: For this investigative assignment, you will receive detailed the details of an unknown solid and liquid using different measurement techniques. Record your data

measurements in the lab report below. You will submit your completed report [HERE](#) after you are [done](#) [with](#) [the](#) [lab](#) [report](#).

It was determined that your solid was "sugar" and liquid was "oil" (olive or almond oil).

Part 1: Mass and Volume Measurements

Enter measured values for the following in the table provided below:

- Mass of solid
- Mass of liquid
- Volume of solid
- Volume of liquid

Enter calculated values, like Density, in the Density Measurements Table

Directions: Fill in the table with the calculated values and liquid and solid values provided, all measurements are in grams and in milliliters (mL) that are given.

Expected: I think the solid would have a higher density than the liquid.

I think that if the unknown liquid was water, it would be a liquid because the density of water is 1.0 g/mL and the density of the solid is 1.2 g/mL.

Part 1: is the mass of the unknown solid, the volume of the solid.

Part 2: is the mass of the unknown liquid, the volume of the liquid.

Directions: Fill in the mass of the solid and liquid.

• Solid mass (g)

• Liquid mass (g)

• Volume (mL)

Flvs Chemistry Segment Exam

Vera Liondas, Graham Ware



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web good question the answer to a problem like this is usually written as the square root of the number for example if we had an equation like $x^2 = 57$ where the square root of 57 would be the value for x then it is normally written $\sqrt{57}$ use the quadratic formula to solve the equation $0 = -7q^2 + 2q + 9$ now